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Excitation of Pr³⁺ ions in alkaline-earth fluorides

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Abstract. Emission and excitation spectra of CaF_2 , SrF_2 , BaF_2 doped by PrF_3 were measured in vacuum ultraviolet region at 11 K. Emission spectra of several 5d-4f bands of CaF_2 and BaF_2 show evident vibronic structure with apparent line separated by 400 cm⁻¹ from the main zero phonon line. No such structure was observed in SrF_2 -Pr. The absence of vibronic structure in SrF_2 is assumed to be caused by the closeness of a local vibronic frequency to the lattice phonon one.

Excitation spectra of 4f-4f and 5d-4f transitions are substantially different pointing on to the different excitation mechanisms of both groups of lines. The 4f-4f lines show an intense excitation peak adjacent to the exciton peak. This infers that excitons created near to \Pr^{3+} ions can feed the 4f-4f transitions. The efficiency of excitation of the 4f-4f and 5d-4f emission bands is quite low in the region of band-to-band transitions. An apparent peak at 19 eV in the excitation spectrum of the BaF₂-Pr 5d-4f emission (230 nm) coincides well with that of the crossluminescence. No similar peaks were observed in CaF_2 or SrF_2 excitation spectra. We conclude that the main channel of energy transfer from the host to Pr^{3+} ions is realized resonantly through crossluminescence in BaF_2 -Pr.

1. Introduction

Crystals doped by Pr^{3+} ions could be considered as perspective materials for scintillator or quantum cutting applications [1]. The mechanism of energy transfer from a host to Pr ions still remains one of the unsolved questions. The spectroscopic investigation in the vacuum ultraviolet region is one of the most powerful methods for studying the mechanisms of energy transfer.

After excitation by vacuum ultraviolet photons of Pr³⁺ ion doped into alkaline-earth fluorides two systems of lines are emitted. Ultraviolet bands observed in the region 215-370 nm belong to 5d-4f allowed interconfigurational transitions, and the sharp lines at 390-900 nm region belong to the 4f-4f parity forbidden intraconfigurational transitions. At low temperatures interconfigurational absorption, excitation and emission bands show rich fine structure due to interaction with vibrational modes [2, 3, 4].

Excitation spectra of BaF₂-Pr in vacuum ultraviolet region were investigated earlier [5]. The authors discussed the possibility of core excitation in the process of energy transfer to praseodymium in barium fluoride. The crossluminescence (or core-valence transitions) occurs from the valence band to the outmost core band where a hole was created by the absorption of a vacuum ultraviolet photon. The crossluminescence of undoped BaF₂ consists of the main band at 220 nm, overlapping strongly with the Pr³⁺ 4f-5d absorption, and a lower-intensity band at

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193 nm [6]. No core-valence transitions (or crossluminescence) were observed in SrF_2 or in CaF_2 where these transitions are energetically unfavourable [6].

The main goal of this paper is the clarification of Pr^{3+} emission and excitation mechanisms by comparative spectroscopic study of the three homologous hosts CaF_2 , SrF_2 , BaF_2 with different concentration of PrF_3 dopants in vacuum ultraviolet region.

2. Experimental

Crystals were grown in vacuum in a graphite crucible by the Stockbarger method. Graphite crucible contains three cylindrical cavities 10 mm in diameter and 80 mm long, which allows to grow three crystals of $\emptyset 10x50$ mm in dimensions with different impurity concentrations at the same time. As the first step a large oxygen-free crystal was grown with addition of CdF₂ as an oxygen scavenger in a simple large volume crucible. Then parts of this crystal were used for growing the doped crystals.

Emission and excitation spectra as well as the emission decay kinetics of CaF_2 , SrF_2 and BaF_2 doped by 0.15 mol. % of PrF_3 were studied in the region of 2-24 eV at 11 K. The measurements were conducted at the SUPERLUMI station of HASYLAB at DESY (Hamburg, Germany) [7]. Additionally some emission spectra were measured at the Institute of Geochemistry using Kr (120 nm) or Xe (147 nm) discharge lamps as excitation sources and a solar-blinde photomultiplier FEU142 attached to a vacuum grating monochromator VM4.

3. Results

The emission spectra due to the 5d-4f Pr^{3+} transitions are shown on figure (Fig.1). The observed emission lines are due to transitions from the lowest 5d-level to different 4f levels of the Pr^{3+} ion, whose energies in CaF_2 are known from [8]. Five 5d levels of Pr^{3+} ion are split by the crystal field in two groups. Usually the 5d-4f emission lines are wider than the 4f-4f lines due to the fact that the outmost d - levels interact with lattice vibration stronger than the inner f - levels. The 5d-4f emission spectrum of CaF_2 -Pr is similar to those measured earlier [3]. Each emission band of CaF_2 and BaF_2 shows a prominent zero phonon line and another strong line of smaller intensity separated by near 400 cm⁻¹. From additional measurements of lower doped crystals the fine structure of 5d-4f emission spectra was evidently observed at 78K in BaF_2 -0.01 % PrF_3 , the structure is less evident in CaF_2 -0.01 % PrF_3 and fully absent in SrF_2 -0.01 % PrF_3 .

The shape of emission spectra are compared in one plot by shifting the spectra by 44300, 45300, 45300 cm⁻¹ for CaF₂, SrF₂, BaF₂, respectively (Fig.2). The spectra of CaF₂ and BaF₂ are similar, while the separation between the sharp lines is slightly smaller in BaF₂. The spectrum of SrF₂ is structureless.

Excitation spectra for the 5d-4f (for the line 5d - 3H_4), 4f-4f (for the line 3P_0 - 3H_4) and exciton emissions are shown in Figure 3. The excitation curves of the 5d-4f and 4f-4f emission are different in the region of direct 5d-4f excitation 5.5-9 eV. Prominent excitation bands were observed below the exciton peaks only. Both spectra are of low intensity and very smooth above the exciton peaks in the range of interband transition (see Fig.3).

4f-4f excitation The excitation spectra of the ${}^{3}P_{0}$ - ${}^{3}H_{4}$ emission line show the peaks near 7 eV (see Fig.3). This band is most intense in CaF₂ less pronounced in SrF₂ and the weakest in BaF₂ crystals. It belongs to the aggregate Pr defects, which possess photon cascade emission [9]. All three crystals show intensive excitation bands right below the exciton peak. Similarly situated wide peaks were observed earlier for the 4f-4f emission of several rare- earth ions in LaF₃ [10].

5d-4f excitation The excitation spectra of the 5d-4f emissions in CaF₂ and SrF₂ show a sharp decrease at the exciton edge energies (see Fig.3). The decrease is less evident in BaF₂. The spectra are of low intensity and almost structureless in the region of the interband transitions. The prominent excitation peak near 19.5 eV was observed for BaF₂, it was absent in CaF₂ and

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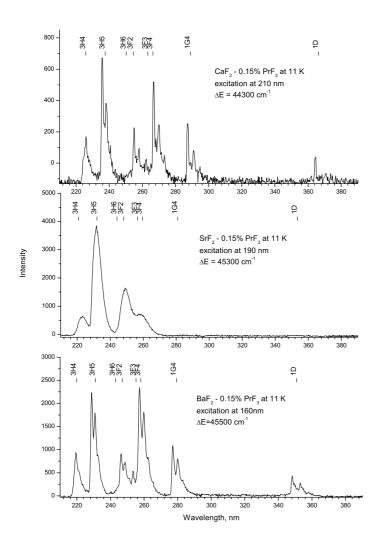


Figure 1. Emission spectra of Pr^{3+} 5d-4f transitions in CaF_2 , SrF_2 , BaF_2 crystals doped by 0.15 molar % of PrF_3 . Positions of each line due to the transitions from the lowest 5d levels to individual 4f levels are shown at the top of each plot.

SrF₂ crystals. The excitation efficiency at 19.5 eV is only twice as low as that in the region of a direct 4f-5d excitation (see Fig.3).

To verify the possibility of energy transfer from the host to a praseodymium ion due to the absorption of a crossluminescence photon by 5d-4f states we compare the emission spectra of undoped and Pr-doped BaF₂ with the absorption spectrum of Pr³⁺ ion in BaF₂ (Fig.4). The absorption is the highest at 200-220 nm where it largely overlaps with 220 nm crossluminescent bands (see Fig.4). With increasing Pr concentration the intensity of the Pr emission lines grows, while the intensity at 210 nm, which is mostly due to the crossluminescence, is continuously decreased partially due to the increasing Pr absorption. The changes in crossluminescence intensity could be monitored via the intensity of the 193 nm band, where the Pr absorption is relatively low. The intensity of the 193 nm band also decreases with increasing Pr concentration (see Fig.4).

4. Discussion

It should be noted that fine structure of spectra was not observed in SrF_2 -0.15 % PrF_3 crystal (see Fig.1). Usually many physical properties changes continuously in the row CaF_2 - SrF_2 - BaF_2 [11]. Therefore the observed dependence is very unusual. The transitions from the 5d to 4f state

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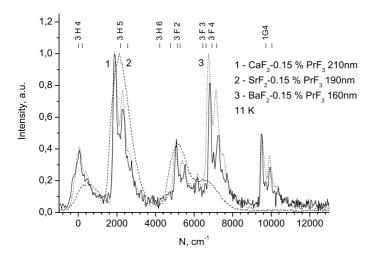


Figure 2. Comparison of the shapes of the emission spectra of Pr^{3+} 5d-4f transitions in CaF_2 , SrF_2 , BaF_2 . The spectra are normalized for the 3H_5 line intensities, while the origin of the wave number scale is placed at the 3H_4 line.

are often assisted by lattice phonons. As a result the optical emission or absorption spectrum contains phonon wing. The emission or absorption spectra of Ce³⁺ in SrF₂ consist of a zerophonon line and a broad phonon wing shifted from line by 400 cm⁻¹ [12]. The frequencies of a longitudinal optic phonon obtained from optical data at 5 K are 484, 397 and 346⁻¹ for CaF₂, SrF₂, BaF₂ respectively [11]. Typical distances between the vibronic lines in absorption spectra were 450, 430, 400 cm⁻¹ for the 0.005 % PrF₃ doped CaF₂, SrF₂, BaF₂ crystals respectively [4]. The spectrum of several individual Pr emission lines caused by transitions from the 5d to ${}^{3}\mathrm{H}_{4}$, ${}^{3}\mathrm{F}_{2}$, ${}^{3}\mathrm{F}_{3}$, ${}^{1}\mathrm{G}$ states in BaF₂ and CaF₂ is shown in Fig.2. Besides the zero-phonon lines the intensive vibrational lines separated by 430 and 390 cm-1 were observed in CaF₂ and BaF₂, respectively. For CaF₂ similar 5d-4f emission spectra of Pr ion were measured with higher resolution earlier [3]. Obviously these phonon lines are due to a local vibration in the complex PrF₈ (see also [4]). The frequencies of these local vibrations are higher than the lattice phonon frequencies in BaF₂, lower then those in CaF₂ and should be almost equal to the phonon frequencies in SrF₂. Therefore the longitudinal optic phonons possessing the energy equal to that of the local vibrations of PrF₈ should wide the Pr emission lines in SrF₂ (see Fig.2). In CaF₂ or BaF₂, the frequencies of lattice phonons and PrF₈ local vibrations are sufficiently different and therefore the structure of emission spectra can be observed.

Excitation spectra of the intraconfigurational 4f-4f transitions of trivalent Nd, Er, Tm, Pr ions doped into several trifluorides were measured in an early paper [10]. Excitation bands on the low-energy wing of exciton bands were observed in all cases. Evidently the excitonic mechanism of the energy transfer to the f-shell has a common character. One can assume that the energy of self-trapped exciton (near 4.5 eV in fluorides) is transferred to the appropriated excited f-levels of rare-earth ions. The exciton energy is not sufficient for the excitation of 4f-5d transitions for all the rare-earth ions except Ce³⁺.

Excitation spectrum for the 255 nm Pr^{3+} emission (transition $5d^{-3}F_{j}$) was measured for BaF_{2-} 0.3 % PrF_{3} earlier [5]. The integrated intensity and the fast component of the 5d luminescence show a rather smooth increase in intensity above 18 eV, that is in the region of excitation of the 5p Ba core states. Authors assume that the energy of the core cation excitations to be transferred to Pr luminescence centres in BaF_{2} [5]. Our measurements show an evident Pr 5d-4f excitation peak which well coincides with the crossluminescence excitation peak in undoped BaF_{2} crystals. The decrease in the crossluminescence intensity accompanying the increase of Pr concentration is clearly seen from comparison of the emission intensity at the wavelength 193 nm, where the

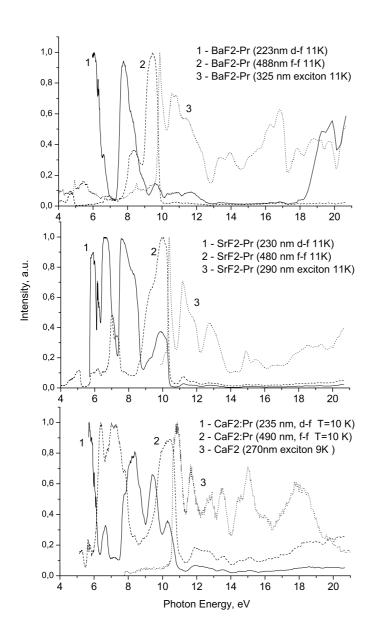


Figure 3. Excitation spectra of Pr 5d-4f and 4f-4f emission bands as well as exciton emission of CaF₂, SrF₂, BaF₂ crystals doped by 0.15 molar % of PrF₃.

Pr absorption is low (see Fig.4). At the level of 0.3 molar % of PrF_3 the crossluminescence is almost completely suppressed. Considering the uniform Pr distribution across the BaF_2 lattice one could estimate the radius of the resonant transfer from a crossluminescent centre to a Pr ion, at which the crossluminescence becomes fully suppressed, as half the distance between Pr ions. The half Pr-Pr distance at the level of 0.3 molar % is near 21.5 Å.

One could estimate this radius also from emission and absorption spectra overlapping. The radius R_c of dipole-dipole energy transfer defined as the distance at which the probability of donor (crossluminescent centre in our case) radiative transitions is equal to the probability of transfer to acceptor (Pr ion), is given by [13]:

$$R_c^6 = \frac{B}{n^4 N_A} \int_0^\infty \frac{f_D(E)\mu_A(E)}{E^4} dE$$

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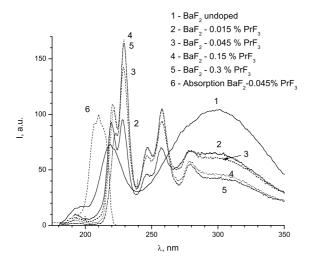


Figure 4. Emission spectra of undoped BaF₂ (curve 1) and BaF₂-PrF₃ (curves 2-5) measured under x-ray excitation at room temperature. Absorption spectrum of BaF₂-0.045 % PrF₃ in arbitrary units is added for comparison (curve 6).

Here n is the refractive index of crystal, the subscript A denotes the acceptor centre and D denotes the donor centre. The term N_A is concentration of acceptor centres (in cm⁻³) and μ_A (E) is their absorption coefficient (in cm⁻¹). The $f_D(E)$ is the emission spectrum of the donor centres, normalised in such a way that $\int\limits_0^\infty f_D(E)dE=1$. Constant $B=\frac{3\hbar^4c^4}{4\pi}=3.7*10^{-20}(eV^4*cm^4)$. From the experimental data we evaluate the radius R_c as 18.3 Å. Thus, both values obtained for the distance of energy transfer from the host to the \Pr^{3+} ion via the crossluminescence are in good agreement.

We can summarize the results supporting the energy transfer to the Pr^{3+} ions by crossluminescence in BaF_2 -Pr as follows:

- sufficient overlap between the crossluminecence and Pr absorption,
- continuous decrease of crossluminescence intensity (at 193 nm) with Pr concentration increasing,
- appearance of the cross luminescence excitation peak near 19 eV in the Pr 5d-4f excitation spectrum.

5. Conclusion

The above-described experimental results prove that the Pr^{3+} 5d-4f emission is efficiently excited via the resonant energy transfer from crossluminescence centres in BaF_2 . The excitation mechanism in CaF_2 and SrF_2 should be investigated additionally.

6. Acknowledgments

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