

Supporting Information

Chemical Framework to Design Linear-like Relaxors toward Capacitive Energy Storage

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Method:

Sample preparation: $(\text{Bi}_{0.35}\text{K}_{0.175}\text{Ba}_{0.3}\text{Na}_{0.175})(\text{Ti}_{0.7}\text{Zr}_{0.3})\text{O}_3$ (BKT-D1), and $(\text{Ba}_{0.45}\text{Bi}_{0.35}\text{Ca}_{0.2})(\text{Ti}_{0.65}\text{Sc}_{0.35})\text{O}_3$ (BT-D2) solid solution ceramics were prepared by the solid-state reaction method. The dried raw powder including Bi_2O_3 (99.9%, 3% excess), Na_2CO_3 (99.8%, 3% excess), K_2CO_3 (99%, 5% excess), BaCO_3 (99.95%), TiO_2 (98%), ZrO_2 (99.99%), Sc_2O_3 (99%) and Nb_2O_5 (99.9%) were weighed according to the stoichiometric ratio. The mixture was then planetary ball-milled with ethanol for 12 h. After drying, the mixed powders were calcined at 800 ~ 850 °C for 2 h. Some of the calcined powders were pressed into pellets and sintered at 1080 ~ 1180 °C for 1~3 h, with the remaining powders used to compensate for the evaporation loss of K, Na, and Bi in a closed crucible. The surface layers of the sintered pellets were removed, and some of the pellets were ground into powders for further analysis.

Phase and crystal structure identification: The room temperature high-energy synchrotron X-ray powder diffraction was performed at the P02.1 beamline of the PETRA III facility by using an X-ray wavelength of 0.2074 Å with a Varex XRD 4343CT 2D detector at a distance from the sample of 2200 mm¹.

Microstructural characterizations: The surface morphologies of ceramics were analyzed using a field-emission scanning electron microscopy (LEO1530, ZEISS SUPRA 55, Oberkochen, Germany). High-resolution bright-field TEM images, grain morphology, and corresponding selected area electron diffraction were conducted on an image-aberration corrected scanning transmission electron microscopy (Titan ETEM G2, Thermo Fisher Scientific, USA) operated at 300 kV. Atomic-scale high-angle dark-field annular (HAADF) STEM images were recorded using a scanning transmission microscopy (JEM-F200, JEOL, Japan) equipped with probe and image aberration-correctors operated at 200 kV. To remove noise, all HAADF-STEM images were Fourier-filtered using a lattice mask. The position and intensity of

atomic columns were determined by fitting with a two-dimensional Gaussian function, and the angle and magnitude of atomic displacements were calculated using a MATLAB script.

Electrical properties measurements: The ferroelectric tester (model aixACCT, TF Analyzer 1000) was used to test the low electric-field bipolar P - E loops on ceramic disks with a diameter of 10 mm and a thickness of 0.5 mm. The tests were conducted at room temperature and a frequency of 1 Hz. The dielectric spectra were measured using a dielectric spectrum measurement system (LDM-500, China), while the electrochemical impedance was measured using a Precision impedance analyzer (Agilent 4294A, USA).

Capacitive energy storage properties measurements:

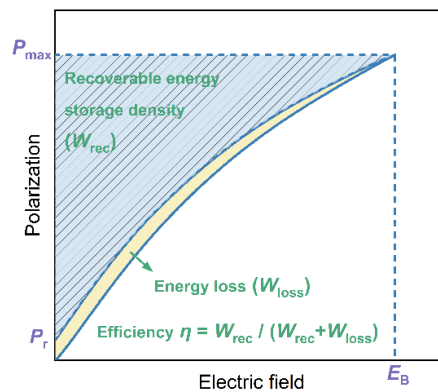
For dielectric capacitors, the total energy storage density (W), energy storage density (W_{rec}) and energy storage efficiency (η) are given by:

$$W = \int_0^{P_{\text{max}}} E dP$$

$$W_{\text{rec}} = \int_{P_r}^{P_{\text{max}}} E dP$$

$$\eta = \frac{W_{\text{rec}}}{W}$$

Where, E is the applied field and P_{max} and P_r are the maximum and remanent polarization, respectively.



Diagrammatic illustration of dielectric energy storage. The shaded area indicates the energy storage density W_{rec} .

The energy storage properties were measured on bulk ceramic samples with thickness of 50-70 μm , and gold electrodes with an area of approximately 0.8 mm^2 . The bulk ceramic samples were polished under water to thicknesses of about 50-70 μm using lapping films (3M) coated with successively finer aluminium oxide particles. The final polish was done using a slurry of 0.05- μm colloidal silica in an alkaline suspension. A gold electrode was evaporated onto one face of the bulk ceramics, serving as the bottom electrode. Gold point-electrodes with an approximate area of about 0.8 mm^2 (about 1.0 mm in diameter) were deposited on the top side of the pellets to serve as the top electrode. The ferroelectric tester from Radiant Technologies Inc., USA was used to test the unipolar P - E loops at various electric fields, frequencies, and temperatures in silicone oil. The discharge characteristics, with an overdamped value of 51 Ω , were measured using Tongguo (TG) technology (CFD-003, China). To assess cyclic stability, the samples were subjected to a cyclic electric field at 10 kHz. After cycling, the energy-storage properties were determined by measuring the unipolar P - E loops of the fatigued samples under 40 kV mm^{-1} for BKT-D1 and 50 kV mm^{-1} for BT-D2 at 10 Hz.

Total scattering measurements: The neutron total scattering data were collected at room temperature using the Nanoscale-Ordered Materials Diffractometer (NOMAD) at the Spallation Neutron Source (SNS), Oak Ridge National Laboratory. The powder sample was loaded in quartz capillaries. The data, which had a Q_{max} of 35 $\text{\AA}^{-1}$, were processed using the ADDIE software to correct for attenuation, inelasticity effects, multiple scattering, and background. The pair distribution function $G(r)$ was then determined by Fourier transforming the corrected total scattering structure function $S(Q)$. To assess the instrument resolution, total scattering data for NIST Si SRM powder were also measured. The calculated total-scattering data were adjusted for the instrument resolution in both reciprocal and real spaces. It should be noted that the

distinct neutron scattering lengths of various elements (Table S5) allowed for the differentiation of element-specific structural information.

Big-box refinement based on the Reverse Monte Carlo method: The Bragg Rietveld refinement was performed with GSAS-II software using the neutron Bragg data to obtain the long-range average crystal structure parameters (Table S1). Reverse Monte Carlo (RMC) fitting was performed on the RMCprofile software^{2,3}. The obtained long-range average cubic structure was expanded to a 20×20×20 super-cell (about 80×80×80 Å³) with containing 40000 atoms (including 2800 Bi atoms, 1400 K atoms, 1400 Na atoms, 2400 Ba atoms, 5600 Ti atoms, 2400 Zr atoms, and 24000 O atoms). The *A/B*-site atoms were randomly distributed initially. In the RMC simulations, the neutron $G(r)$, $S(Q)$, and Bragg data, in total three data, were simultaneously fitted under bond valence sum constraints and coordination constraints. *A*-site and *B*-site atoms were allowed to swap. During the RMC simulation process, RMCprofile automatically optimizes the weights among the fitting data. Each RMC modeling was run to generate more than 10³ moves per atom to get converged. The details of the analysis of the polar atomic displacements and unit-cell polar vectors from the refined 3D atom configurations can be found in our previous works⁴⁻⁵⁶. To get the statistical analysis results, 20 refined 3D atom configurations were merged together.

EXAFS. Room temperature Bi L_{III}-edge EXAFS spectra were collected in transmission mode at the XAFS beamline of ELETTRA Sincrotrone Trieste during the experiment no. 20230135. The sample for EXAFS was prepared by mixing and pelletizing the sample powder with boron nitride powder. The amount of sample powder was chosen to have an absorption edge jump $\Delta\mu x \sim 1$. The EXAFS signal was extracted by using Athena and Artemis software^{7,8}, and FEFF8⁹ was employed to calculate the Bi-O (and Bi-(Ti)) scattering amplitudes and phases shift, using BiTiO₃ as reference. An in-house software based on FEFFIT code was developed to perform the best-fitting simulation of the EXAFS Fourier transform (FT) between ~ 0.5 and 3.5 Å. The preliminary simulations demonstrate that two distinct Bi-O distributions (+

one longer Bi-(Ti) distance distribution) are needed to fit the FT structure (Fig. S16). Various best-fitting simulations with different FT intervals and k-weight were performed, and then averaged (Fig. S17).

Calculation of electronic charge density based on the density functional theory.

The electronic charge density from the respective clusters in a refined 3D atomic configuration based on the density functional theory calculations (Fig. 4j and Fig. S19). The first-principles calculations are based on density functional theory within the Perdew-Burke-Ernzerhof approximation¹⁰, as implemented in the Vienna ab initio simulation package¹¹. Interactions between ion cores and valence electrons are described by the projector augmented wave method¹². Plane waves with a kinetic energy cutoff of 500 eV are used as the basis set.

Mechanical performance measurements. The Vickers hardness values of sample were tested by wolpert wilson instruments (401MVD, China). The applied load for each indentation process was set at 9.8 N with a dwell time of 15 s.

Supplementary Figures and Tables

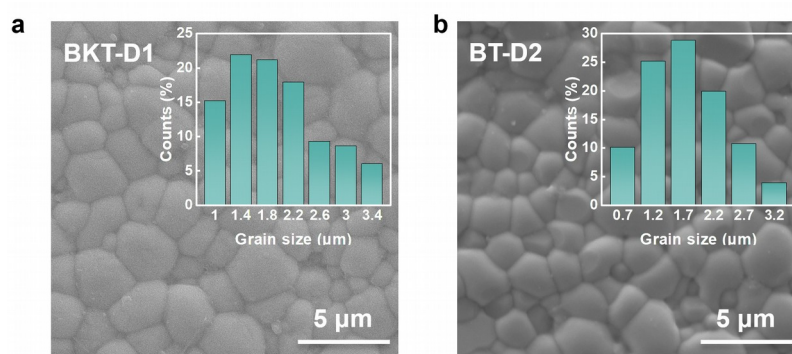


Figure S1. SEM images of BKT-D1 and BT-D2 ceramics, the insets indicate the corresponding statistical distribution of grain size.

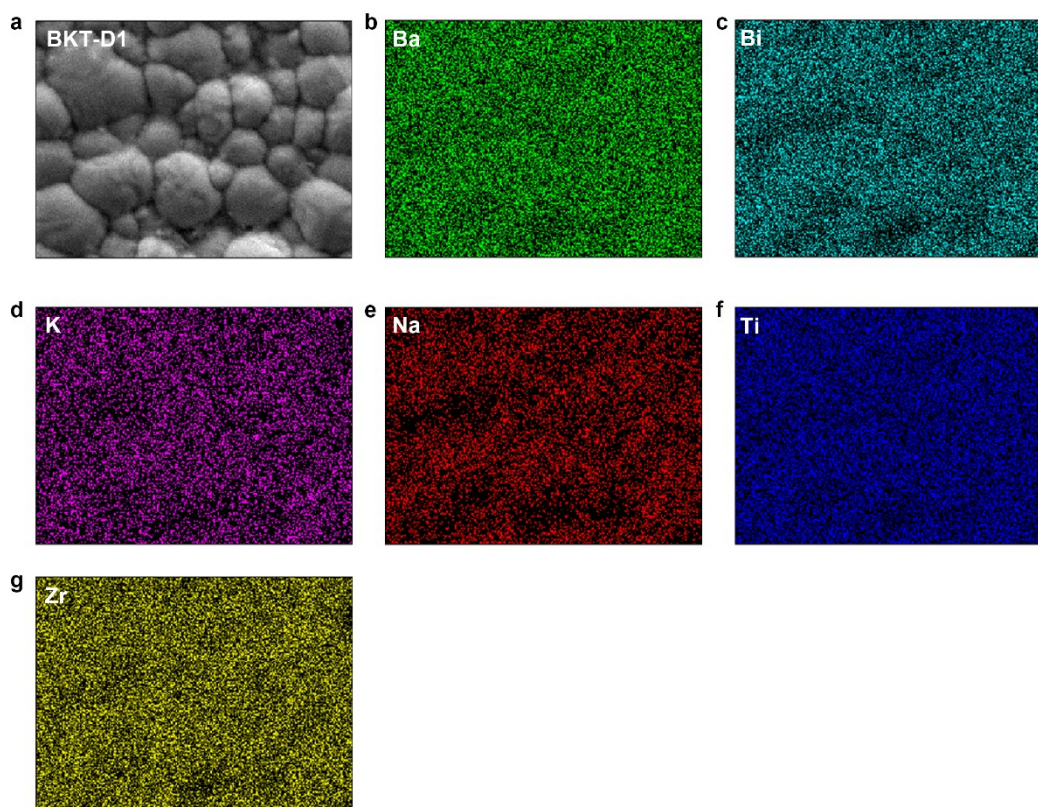


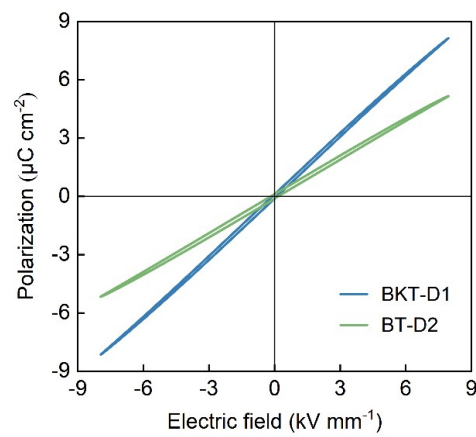
Figure S2. (a) SEM surface morphology and (b-g) the corresponding element mapping of BKT-D1.

Table S1. Crystal structure parameters of BKT-D1 obtained from Rietveld refinement.

BKT-D1					
Space group: $Pm-3m$					
Cell parameters: $a=b=c=4.0256 \text{ \AA}$, $\alpha=\beta=\gamma=90^\circ$					
atom type	occupancy	atomic coordinate			Uiso
		x	y	z	
Bi	0.35	0	0	0	0.070
Na	0.175	0	0	0	0.070
K	0.175	0	0	0	0.070
Ba	0.3	0	0	0	0.070
Ti	0.7	0.5	0.5	0.5	0.009
Zr	0.3	0.5	0.5	0.5	0.009
O	1	0.5	0	0.5	0.020
$R_w = 6.5\%$, $GOF = 2.0$					

Table S2. Crystal structure parameters of BT-D2 obtained from Rietveld refinement.

BT-D2					
Space group: $Pm-3m$					
Cell parameters: $a=b=c=4.0213 \text{ \AA}$, $\alpha=\beta=\gamma=90^\circ$					
atom type	occupancy	atomic coordinate			Uiso
		x	y	z	
Ba	0.45	0	0	0	0.050
Ca	0.2	0	0	0	0.050
Bi	0.35	0	0	0	0.050
Ti	0.65	0.5	0.5	0.5	0.005
Sc	0.35	0.5	0.5	0.5	0.005
O	1	0.5	0	0.5	0.022
$R_w = 7.6\%$, $GOF = 2.7$					

**Figure S3.** Bipolar P - E loops of BKT-D1, and BT-D2 ceramics measured at the lower electric field of 8 kV mm^{-1} at 1 Hz at room temperature.

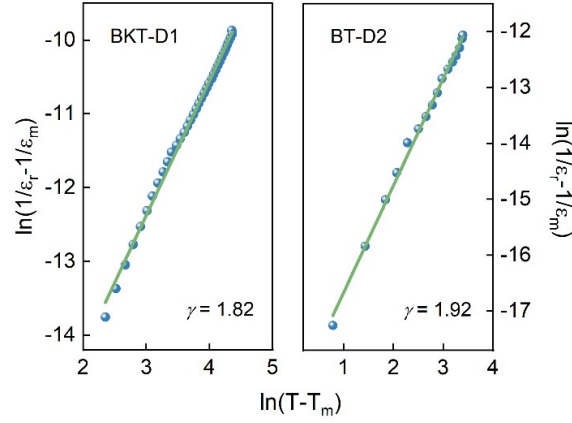


Figure S4. Characterization of the diffuseness parameter γ based on the modified Curie-Weiss distribution.

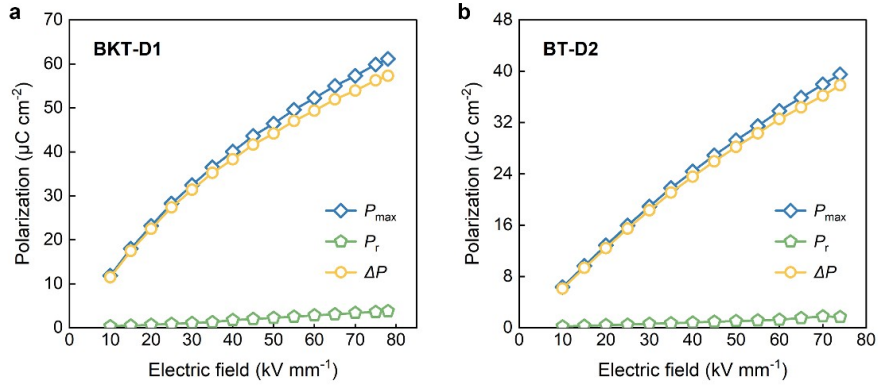


Figure S5. The extracted P_r , P_m , and ΔP as function of the electric field. (a) BKT-D1 and (b) BT-D2.

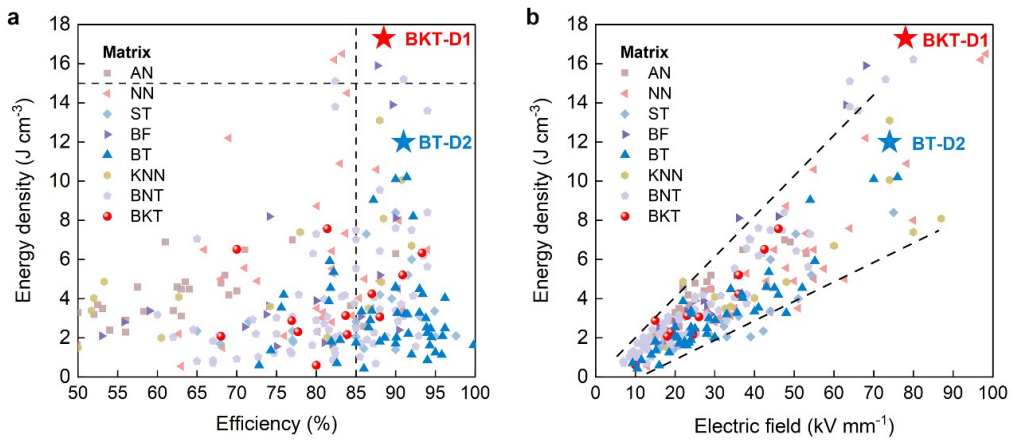


Figure S6. Comparison of W_{rec} versus (a) efficiency (η), (b) E_b between BKT-D1 and

BT-D2 bulk ceramics, and other reported lead-free bulk ceramics.

Table S3. A comparison of W_{rec} and η between BKT-D1 and other reported BKT-based lead-free bulk ceramics.

Composition	W_{rec} (J cm ⁻³)	η (%)	Reference
0.5BKT–0.5ST	0.6	80	Ref. ¹³
BKT-0.06LMT	2.08	68	Ref. ¹⁴
SBKT-2wt.% Er₂O₃	2.17	83.9	Ref. ¹⁵
0.7BKT-0.3ST	2.31	77.7	Ref. ¹⁶
0.76 BKT-0.24BF	2.88	76.9	Ref. ¹⁷
0.95(0.5BKT–0.5ST)– 0.05BZT	3.07	88	Ref. ¹⁸
BKT-0.15BMN	3.14	83.7	Ref. ¹⁹
0.96(0.6BKT-0.4SBT)- 0.04BMH	4.25	87	Ref. ²⁰
0.6 (0.75BKT-0.25BF) -0.4SBT	5.21	90.87	Ref. ²¹
0.5(0.7BKT-0.25BF)-0.5CT	6.34	93.32	Ref. ²²
0.84(0.6BKT-0.4BF)- 0.16NSN	6.52	70	Ref. ²³
0.6BKT-0.3BT-0.1NN	7.57	81.4	Ref. ²⁴
BKT-D1	17.3	88.5	This work

Table S4. A comparison of W_{rec} and η between BT-D2 and other reported BT-based lead-free bulk ceramics.

Composition	W_{rec} (J cm ⁻³)	η (%)	Reference
BTMN	0.43	86	Ref. ²⁵
BCZT	0.59	72.8	Ref. ²⁶
0.91BT-0.09BY	0.71	82.6	Ref. ²⁷
0.9BT-0.1BY-MWS	0.86	94	Ref. ²⁸
0.9BT-0.1BMN	1.13	95.8	Ref. ²⁹
BSZT	1.15	92	Ref. ³⁰
0.85BT-0.15BMZ	1.25	95	Ref. ³¹
0.92(0.65BT-0.35BNT)- 0.08SYN	1.36	74.3	Ref. ³²
0.8BT-0.15BMZ	1.61	94.3	Ref. ³³

0.88BST-0.12BZN	1.62	99.8	Ref. ³⁴
Mn-doped 0.9BT-0.1BMN	1.7	90	Ref. ³⁵
BSCT	1.75	85	Ref. ³⁶
0.88BT-0.12BMT	1.81	88	Ref. ³⁷
CuO doped BSCTZ	1.81	81.9	Ref. ³⁸
0.875BT-0.125BMN	1.89	83	Ref. ³⁹
BSTM-2 wt. % MgO	2.01	88.6	Ref. ⁴⁰
0.9BBNT-0.1SSN	2.02	90.1	Ref. ⁴¹
0.96BT-0.04KN	2.03	94.5	Ref. ⁴²
0.88BT-0.12BNN	2.09	95.9	Ref. ⁴³
0.85BT-0.15BZS	2.21	91.6	Ref. ⁴⁴
0.88BT-0.12BMS	2.25	94	Ref. ⁴⁵
0.92BT-0.08BZZ	2.47	94.4	Ref. ⁴⁶
0.84BT-0.16BNZ	2.49	96.2	Ref. ⁴⁷
0.85BCT-0.15NNT	2.53	86.6	Ref. ⁴⁸
0.9BT-0.1BNS	2.53	93.9	Ref. ⁴⁹
0.85BT-0.15BMZ	2.9	86.8	Ref. ⁵⁰
0.88BT-0.12BLZ	3	93.8	Ref. ⁵¹
Gd-doped 0.7BT-0.3SBT	3.2	91.5	Ref. ⁵²
0.6BZT-0.4BNT	3.22	91.2	Ref. ⁵³
0.92BT-0.08LZH	3.27	90.22	Ref. ⁵⁴
0.88BT-0.12BMT	3.28	93	Ref. ⁵⁵
0.9BST-0.1BMN	3.34	85.7	Ref. ⁵⁶
0.9BT-0.1BMH	3.38	87	Ref. ⁵⁷
0.65BLT-0.35SBT	3.54	75.6	Ref. ⁵⁸
0.95BT-0.05BMT-Sm	3.86	81.6	Ref. ⁵⁹
0.65BT-0.35(SBT-BMZ)	4.03	96.2	Ref. ⁶⁰
BSBNT	4.2	75.9	Ref. ⁶¹
0.9[0.9BCT-0.1BLZ]-0.1BNT	4.23	93.4	Ref. ⁶²
0.6BT-0.4BMT	4.49	93	Ref. ⁶³
0.93BST-0.07BMN	4.55	81.8	Ref. ⁶⁴
BT-0.06BMN	4.55	90	Ref. ⁶⁵
0.4BS-0.55BT-0.05NN	4.6	90	Ref. ⁶⁶
0.3BCZT-0.7ST-0.3mol% MnO₂	5.36	82.2	Ref. ⁶⁷
0.93BSZT-0.07BMN	5.92	81.7	Ref. ⁶⁸
BNBSCT-2mol% PbO	8.2	92.2	Ref. ⁶⁹
0.85(0.8BT-0.2BNT)-0.15CZ	9.04	87.2	Ref. ⁷⁰
BBT	10.1	90	Ref. ⁷¹
0.8(0.97BT-0.3NN)-0.2BZZ	10.2	91.4	Ref. ⁷²
BT-D2	12	91	This work

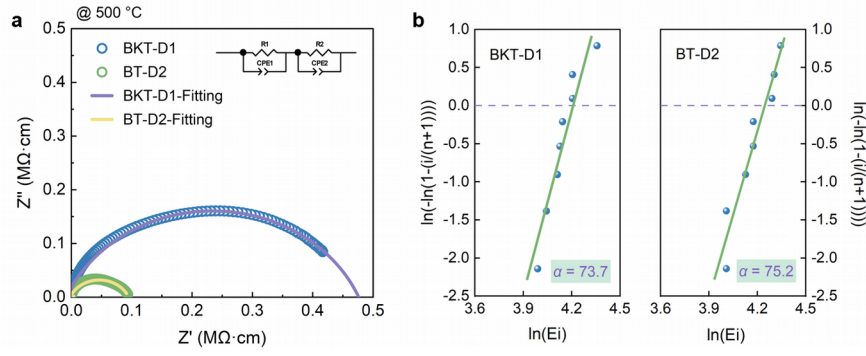


Figure S7. (a) The impedance fitting spectrum at 500 °C of BKT-D1 and BT-D2. The inset shows the calculated band gap. (c) Fitting the Weibull distribution of BKT-D1 and BT-D2.

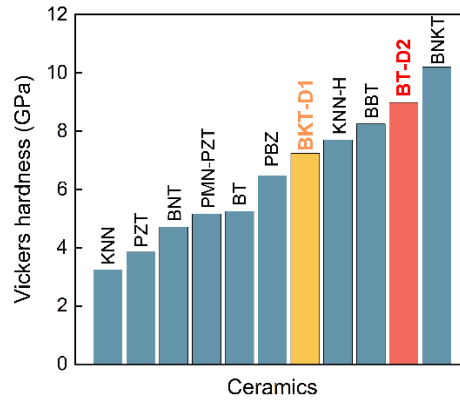


Figure S8. A comparison of Vickers hardness (H_v) between the designed BKT-D1, BT-D2 and some representative perovskite ceramics.

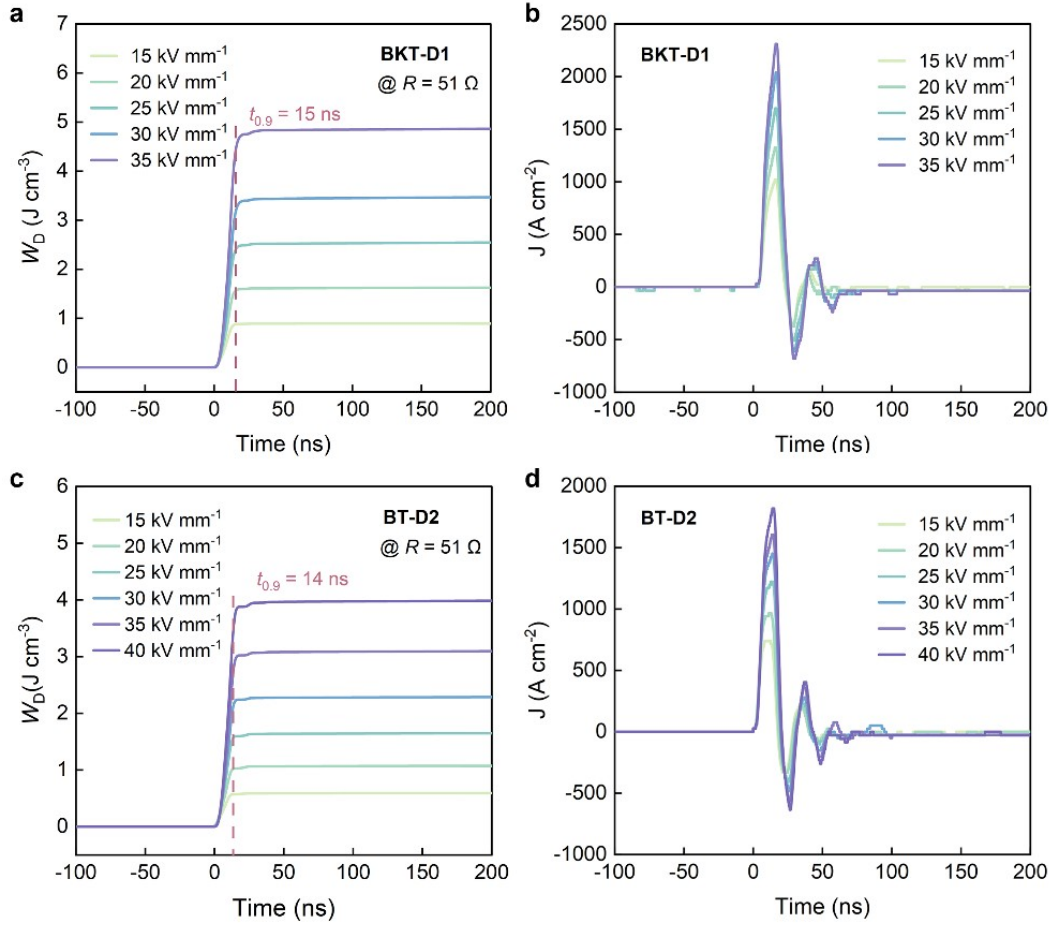


Figure S9. Charge-discharge performance of BKT-D1 and BT-D2. (a) and (c) overdamped discharge energy density. (b) and (d) underdamped discharge waveforms. Note that the applied maximum electric field values during the charge-discharge performance tests are relatively lower compared to those applied for the unipolar polarization curve tests. This is due to the breakdown of the electrodes, rather than the breakdown of the dielectric ceramics themselves in charge-discharge performance tests. If the electrodes of the charge-discharge instrument are optimized, higher electric field values can be applied.

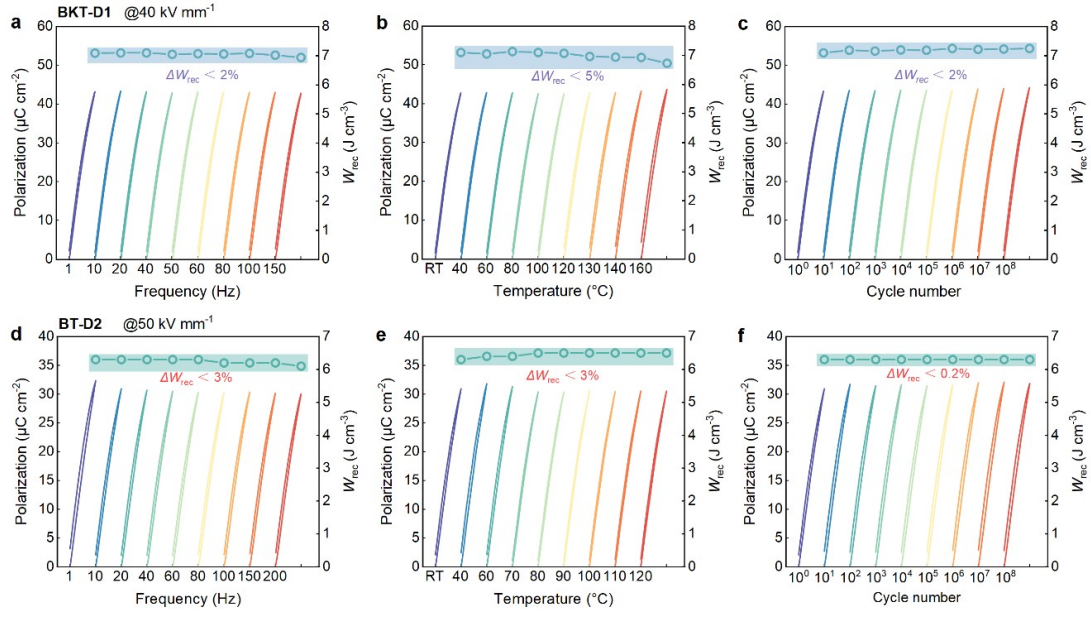


Figure S10. Unipolar P - E loops and the W_{rec} obtained from reliability testing measured at 40 kV mm^{-1} for BKT-D1 and 50 kV mm^{-1} for BT-D2.

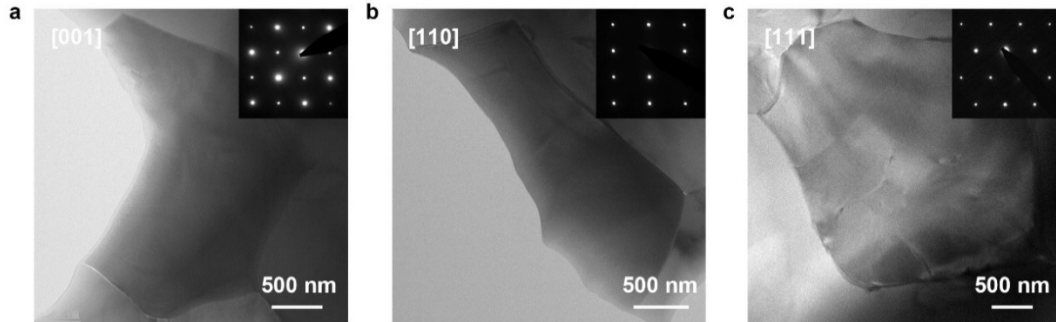


Figure S11. Grain morphology and corresponding SAED captured by high-resolution bright-field TEM images along (a) $[001]_c$, (b) $[110]_c$, (c) $[111]_c$.

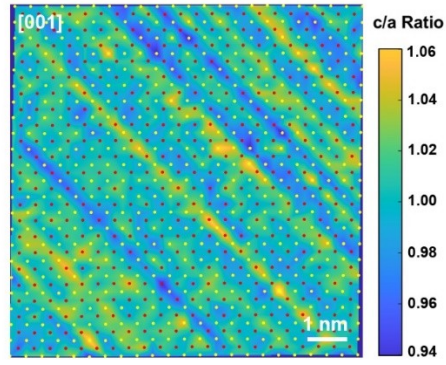


Figure S12. The corresponding lattice distortion (c/a) mapping of Fig. 3b.

Table S5. The neutron scattering lengths of the elements contained in BKT-D1.

Elemen	
t	Neutron scattering lengths (fm)
Bi	8.532
Na	3.63
Ba	5.07
K	3.67
Ti	-3.438
Zr	7.16
O	5.803

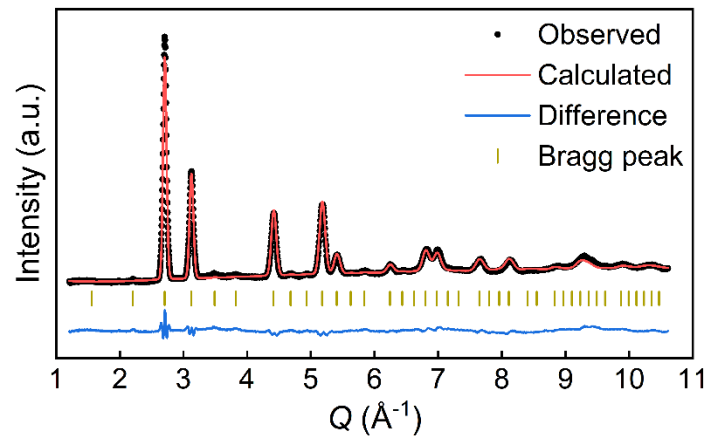


Figure S13. The RMC fitting of neutron Bragg data for BKT-D1.

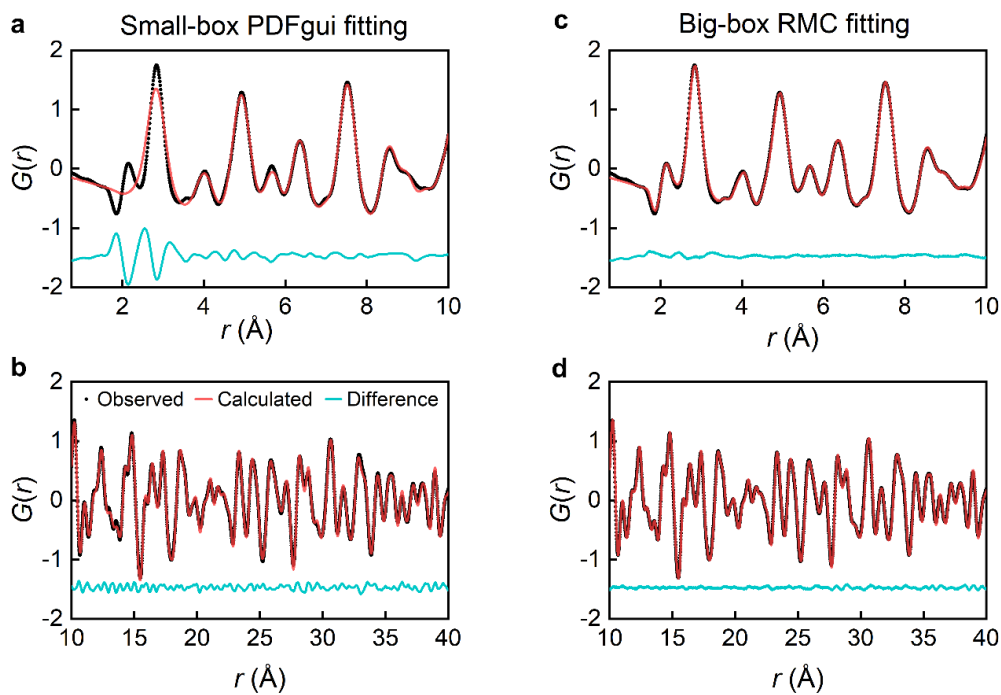


Figure S14. Comparison of the fitting results of neutron PDF of BKT-D1 between the small-box method by using a cubic model (a,b) and the big-box RMC method (c,d).

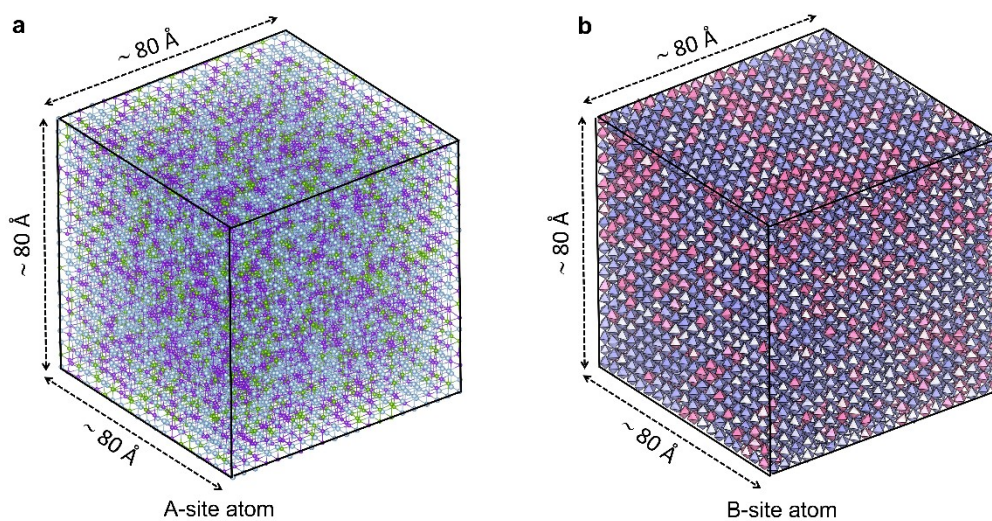


Figure S15. Atomic distribution of BKT-D1 in a refined 3D atom configuration. (a) A-site atomic distribution. The purple, green, and white atoms indicate the Bi, Na, K/Ba, respectively. (b) BO₆ distribution. The pink and lila indicate the TiO₆, and ZrO₆, respectively.

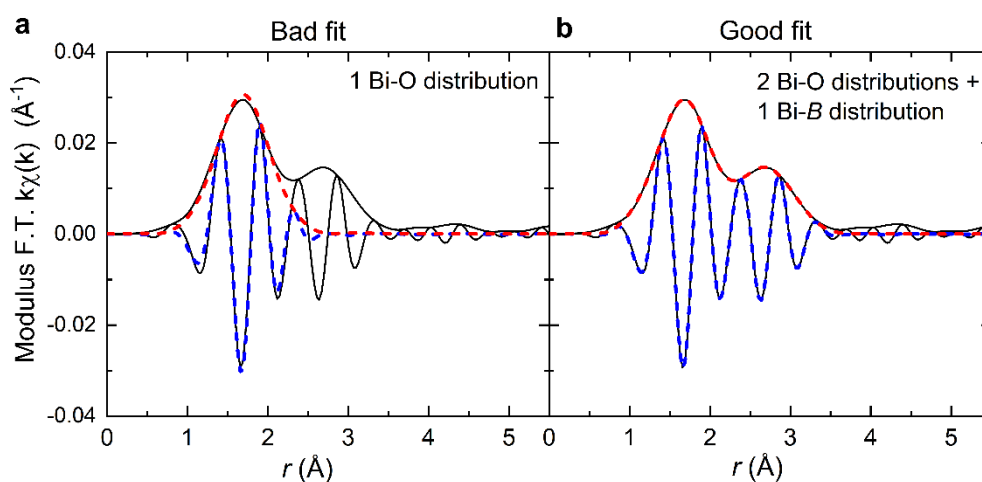


Figure S16. Example of best fitting-simulation of the Bi-L_{III} EXAFS FT (here $k=2-14$ Å⁻¹, k , Gauss) between $R=0.5-3.5$ Å by using: (a) one Bi-O distance distribution and (b) two Bi-O + one longer Bi-(Ti) distance distributions. It can be observed as all three distributions must be included to fit the FT structure.

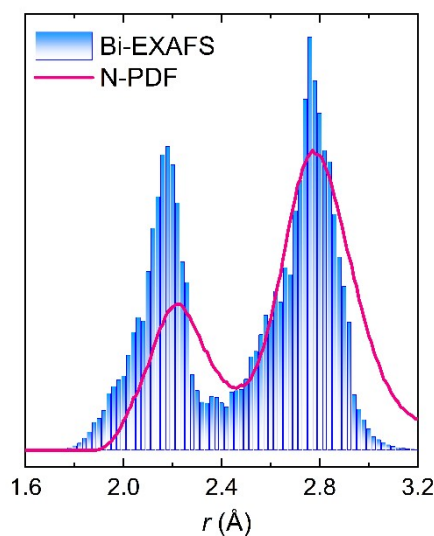


Figure S17. Bi-O distance distribution obtained by Bi L_{III} EXAFS best-fitting simulation and its comparison with partial Bi-O PDFs obtained from neutron total scattering.

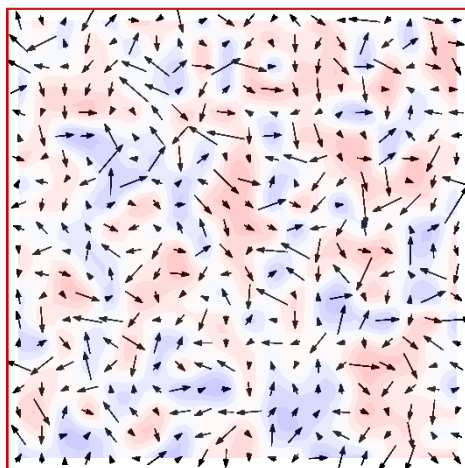


Figure S18. 2D displacement distribution obtained from one of the refined 3D atomic configurations viewed along $[001]_c$. The vector indicates the vector of the *B*-site position atom relative to the center of its four nearest neighboring *A*-site positions, which is an average of 10 unit cells (with thickness about 4 nm) along the $[001]_c$ direction.

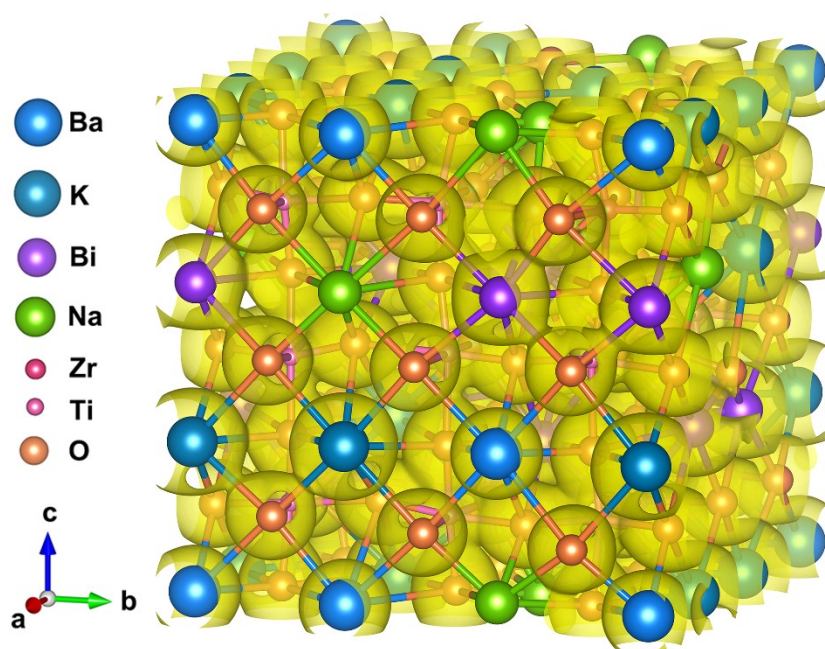


Figure S19. The calculated electronic charge density from the respective clusters in a refined 3D atomic configuration based on the density functional theory calculations, showing the valence electron localization isosurfaces visualized for a value of $0.2 e \text{ \AA}^{-3}$.

Supplementary **References**

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