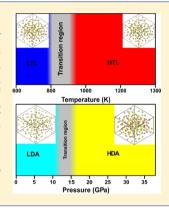
Temperature- and Pressure-Induced Polyamorphic Transitions in AuCuSi Alloy

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Supporting Information

ABSTRACT: Temperature-induced liquid-liquid phase transition (LLPT) and pressureinduced amorphous-amorphous phase transition (AAPT) have never been simultaneously reported in any single metallic system. In an Au₅₅Cu₂₅Si₂₀ alloy, however, we discovered a temperature-induced LLPT by detecting "reversible λ -anomalies" of the thermal expansion coefficient between two liquid states at ambient pressure, while a pressure-induced AAPT in Au₅₅Cu₂₅Si₂₀ metallic glass (MG) occurs upon compression at ambient temperature. Both LLPT and AAPT are reversible with a hysteresis in temperature and pressure, respectively. Using molecular dynamics simulations and synchrotron X-ray techniques, we elucidate structural differences in both low- and high-pressure Au₅₅Cu₂₅Si₂₀ MG phases and low- and hightemperature Au₅₅Cu₂₅Si₂₀ liquid phases. Electronic transfer between Si and Au or/and Cu atoms occurs in both temperature-induced LLPT and pressure-induced AAPT in the Au₅₅Cu₂₅Si₂₀ alloy.



1. INTRODUCTION

Phase transitions in materials are the subject of great interest because of their cardinal essentialness in earth and planetary sciences, physics, chemistry, and materials science multidisciplines. 1-4 The polyamorphic transition induced by temperature and/or pressure, defined as a transition between two different disordered phases (without long range translation and orientation symmetries and with the same composition) in disordered materials, for example, liquids and amorphous materials, is often more complex as compared to the polymorphic transition in crystalline materials. 5-8 The temperature-induced liquid-liquid phase transitions (LLPT) have been studied both experimentally and theoretically for a few non-metallic substances, for example, supercooled liquid $Si, {}^{9-11}$ P, 12,13 C, 14 $SiO_2, {}^{15,16}$ $Al_2O_3 - Y_2O_3, {}^{17,18}$ $H_2O, {}^{1,19-21}$ and triphenyl phosphite, 22 while the pressure-induced amorphous-amorphous phase transitions (AAPT) have also been reported in tetrahedral 4 and 5 coordinated nonmetallic glasses with directional bonding and an open local environment.^{23–27} Simultaneously, temperature- and pressure-induced polyamorphic transitions in metallic systems have not been reported yet. Here, we report unambiguous experimental evidence for temperature-induced LLPT and pressure-induced AAPT in one alloy of a composition of Au₅₅Cu₂₅Si₂₀. The temperature-induced LLPT occurs in a temperature range of 700-900 K at ambient pressure, characterized by the "reversible λ -anomalies" of thermal expansion coefficient between two liquid phases detected from in situ hightemperature X-ray diffraction (XRD), a broad dip in heat flow and a peak in heat capacity from differential scanning calorimetric (DSC). Subsequently, different structures and thermal parameters are observed for the solid Au₅₅Cu₂₅Si₂₀ samples quenched from low- and high-temperature Au₅₅Cu₂₅Si₂₀ liquid phases. Furthermore, the pressure-induced AAPT in the Au₅₅Cu₂₅Si₂₀ metallic glass (MG) occurs upon compression from ~13.1 to ~15.8 GPa at ambient temperature. Both LLPT and AAPT are reversible. Our ab initio molecular dynamics (AIMD) simulations reproduce the experiment results by revealing the polyamorphic phase transitions in the Au₅₅Cu₂₅Si₂₀ alloy. This finding will trigger more studies in deeply understanding phase transitions in disordered materials, especially metallic systems.

Concerning metallic liquids, a possible metal-nonmetal transition in liquid Na,28 a tetrahedral clustering of nearest neighbors in liquid Li at pressure 150 GPa²⁹ were proposed. For several monoatomic metallic liquids (i.e., Al, Zn, Sn and In), Lou et al.³⁰ found that the average interatomic distance between the center atom and those on the first shell contracts with increasing temperature. Temperature-induced LLPTs above their liquidus temperatures were implied in La₅₀Al₃₅Ni₁₅

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alloy by ²⁷Al nuclear magnetic resonance measurements, ³¹ in Rb, 32 Ce, 33 and Ga³⁴ by high-temperature synchrotron radiation-based XRD experiments and AIMD simulations. A LLPT in the undercooled liquid region was indicated in Zr_{41.2}Ti_{13.8}Cu_{12.5}Ni₁₀Be_{22.5} alloy,³⁵ which was accompanied by changes in the local atomic structure and viscosity but without a detectable density anomaly. LLPTs are sometimes associated with a dynamical crossover with changes in the kinetic fragility. 15,35-38 When observed upon cooling, the liquid evolves from a high-temperature fragile phase with a steep temperature dependence of viscosity and structural relaxation time to a strong phase less affected by temperature changes. Metallic glasses, exhibiting many excellent properties, have attracted considerable attention, which are often densely compacted. Although pressure-induced AAPTs were recently reported in a few f-electron containing (Ce-, Pr-, Yd-based) MGs³⁹⁻⁴⁸ and one case in CaAl MGs,^{49,50} it is still challenging to provide more evidence for pressure-induced AAPT in non-felectron containing MGs.⁵¹

2. METHODS

Au₅₅Cu₂₅Si₂₀ alloy ingots were prepared by arc melting constituent elements with purity higher than 99.99 at. % in a Zr-gettered argon atmosphere. Each ingot was re-melted four times to improve the chemical homogeneity. The liquidus temperature of Au₅₅Cu₂₅Si₂₀ alloy is 654 K. Ribbon Au₅₅Cu₂₅Si₂₀ alloys were solidified in a melt-spun facility from the liquid at two temperatures, one above 900 K and the other below 750 K, which were measured by using a pyrometer with a uncertainty of about 30-50 K. Au₅₅Cu₂₅Si₂₀ alloy ingot was put into narrow tubes with diameter of 0.7 mm and thickness around 0.01 mm. To make the tubes completely evacuated, vacuum was created up to 8.6×10^{-4} Pa and directly sealed. A heating furnace with the heating rate of 10 K/min was used to heat the sealed tubes. During heating, the XRD patterns were gathered, with the beam dimension of 0.3 × 0.3 mm² and wavelength of 0.117 Å via a PerkinElmer 1621 Si detector having 2048 \times 2048 pixels and 200 \times 200 μ m² pixel size. The experiment was performed at sector 11-ID-C of advanced photon source (APS), Argonne National Laboratory, USA. The distance from the detector to the sample was 400.893 mm and for the normalization, total incident flux was recorded. For each XRD pattern, 0.5 acquisition time was set, and to get precise results, 10 XRD patterns were added to get a single XRD file. After measurements, the sample still retained metallic shinning behavior. For proper removal of background from original data, the XRD pattern of an empty tube with same thickness and diameter was recorded. Diffraction intensity I(q) patterns were incorporated under the software package FIT2D⁵⁴ after properly subtracting the background. Afterwards by using PDFgetX3,55 we have taken out the contribution of the self-absorption, polarization, fluorescence absorption and Compton scattering from the resultant data, to obtain the total structure factor S(q).

In situ high-temperature small-angle X-ray scattering (SAXS) experiments were also performed at APS, beamline sector (1-ID). The round process of heating to 1174 K and then cooling to 673 K was carried out at a rate of 10 K/min. The beam size was about 50 μ m vertically and about 200 μ m horizontally with an energy of about 110 keV. The SAXS signals in transmission mode were recorded by a PIXIRAD CdTe photon counting detector located about 6 m away from the sample. The exposure time was 1 s, and 10 images were

summed for each data set (about 1.7 K temperature range). The direct-beam X-ray intensities before and after the specimen were recorded, by which the transmission could be calculated. The dark- and bright-field background subtraction and absolute scaling to a glassy carbon standard were also corrected to get the absolute SAXS intensity of the sample at each temperature.

The heat flow and heat capacity $C_p(T)$ data of liquid $Au_{55}Cu_{25}Si_{20}$ alloy using aluminum pans were determined by DSC in PerkinElmer DSC-7 in a high-purity Ar atmosphere. Both heating and cooling heat flow measurements were conducted using a heating rate of 20 K/min, while for heat capacity measurements, a 5 K step heating method was used, with a heating rate of 5 K/min and isothermal holding time (300 s) after each interval. The heat capacity of liquid $Au_{55}Cu_{25}Si_{20}$ alloy was determined by comparing the alloy heat flow with reference to standard sapphire and empty aluminum pan by using the equation described in literature. ⁵⁶

A Mao-Bell-type diamond anvil cell (DAC) with a culet 300 μ m in diameter was used for Au₅₅Cu₂₅Si₂₀⁵⁷ MG for the in situ high-pressure angle-dispersive XRD experiment. The sample chamber was about ~100 μ m in diameter drilled in a preindented T301 stainless steel gasket. The sample was loaded into the chamber along with ruby to perform the pressure calibration. The helium was used as a pressure-transmitting medium. The in situ high-pressure XRD experiments was conducted at beamline P02.2 of PETRAIII with the wavelength of 0.2952 Å.

The first-principles simulations for Au₅₅Cu₂₅Si₂₀ liquid were executed by employing the Vienna ab initio Simulation package (VASP), 58 based on the density functional theory. The canonical NVT ensemble combined with a Nosé–Hoover thermostat to control the temperature ^{59,60} is used for current work. The general gradient approximation in the form described by Perdew-Burke-Ernzerhof is employed for the exchange-correlation energy functional. By using a time step of 3 fs, Newton's equations of motion was extracted via the velocity Verlet algorithm. Only the Γ point was utilized to sample the Brillouin zone of the supercell. The cubic box used contains 200 atoms with periodic boundary conditions. The system was heated and equilibrated at 2000 K to remove the crystalline symmetry, for 8000 AIMD steps, from the initial configurations, and then quenched stepwise to 1500, 1400, 1300, 1200, 1100, 1150, 1000, 900, 850, 800, 750, 700, and 600 K with a cooling rate of 0.1 K/step. At each temperature, the equilibrium volume was maintained during AIMD simulations by keeping the internal pressure approximately zero within about ~1.0 kbar. After adopting the required internal pressure accuracy, the data were run for further 8000 steps for achieving the equilibrium state. The last 2000 steps were used for statistical analyses.

The AIMD simulations for the pressure effect on ${\rm Au}_{55}{\rm Cu}_{25}{\rm Si}_{20}$ MG were also executed by using the Vienna ab initio Simulation package (VASP) by following the same conditions and box size as mentioned above. However, the systems were melted and equilibrated at 2000 K for 8000 steps to remove the memory of the initial configurations and then quenched into steps down to 300 K with a cooling rate of 0.1 K/step. The equilibrium volume was established at 300 K by keeping the internal pressure approximately zero (about ± 1.0 kbar). Similarly, at ambient temperature, pressure was applied by uniformly decreasing the volume of the box. At each pressure point, after getting the required pressure value, the

box was relaxed further for 10 000 steps to achieve an equilibrium state. The last 2000 steps data were used for statistical analyses.

3. RESULTS AND DISCUSSION

Figure 1 illustrates total structure factors S(q) of an $Au_{55}Cu_{25}Si_{20}$ liquid alloy extracted from in situ high-energy

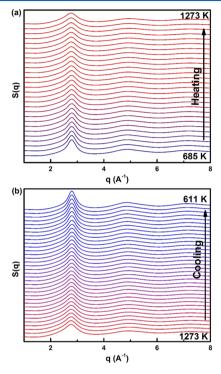


Figure 1. In situ high-energy high-temperature synchrotron XRD patterns. Total structure factors S(q) at various temperatures obtained from in situ high-energy synchrotron XRD measurements for liquid $\mathrm{Au}_{55}\mathrm{Cu}_{25}\mathrm{Si}_{20}$ alloy, (a) during heating from 685 to 1273 K and (b) during cooling from 1273 to 611 K.

XRD measurements over the entire heating (685–1273 K) and cooling (1273-611 K) cycle. Figure 2a shows the first peak position, q_1 , of S(q) as a function of temperature during the entire heating and cooling cycle. During heating, the first peak position shifts to small q values. A deviation from the linearity in the temperature dependence of the peak position occurs at around 770 to 900 K. After that, q_1 shifts linearly again to low q_1 values with increasing temperature up to 1273 K. Upon subsequent cooling, q_1 first follows the heating trace until a sudden decrease at around 770 K, and then increases linearly again below about 700 K. The slopes of q_1 versus T in the temperature range below 770 K and above 900 K during heating are estimated to be $-2.62(3) \times 10^{-5}$ and $-8.13(3) \times 10^{-5}$ 10^{-5} (Å K)⁻¹, respectively, while during cooling, they are $-2.66(3) \times 10^{-5} (\text{Å K})^{-1} \text{ below 700 K and } -8.17(3) \times 10^{-5}$ $(Å K)^{-1}$ above 770 K, revealing that the slope change from the low-temperature range to the high-temperature range in liquid Au₅₅Cu₂₅Si₂₀ alloy during heating is reversible during cooling within the experimental uncertainty, despite a slight shift of the temperature range. The reproducible relationship between q_1 and T during heating and cooling demonstrates that a temperature-induced LLPT likely exists in the liquid Au₅₅Cu₂₅Si₂₀ alloy. To quantitatively evaluate the thermal expansion for liquid Au₅₅Cu₂₅Si₂₀ alloy, the method, that is, the

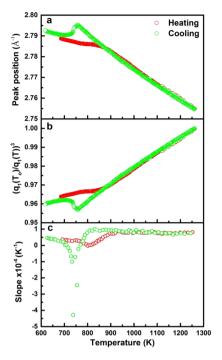


Figure 2. Structural changes. Temperature-dependent structural changes during entire heating and cooling cycle for liquid $\mathrm{Au}_{55}\mathrm{Cu}_{25}\mathrm{Si}_{20}$ alloy, (a) first peak position of S(q), (b) parameter of $(q_1(T_o)/q_1(T))^3$, and (c) slopes of the $(q_1(T_o)/q_1(T))^3$ vs T.

ratio of $(q_1(T_0)/q_1(T))^3$ is approximately proportional to the relative change of sample volume for glassy alloys and also metallic liquids at various temperatures, although the proportional coefficients might be slightly different between glassy alloys and metallic liquids, $^{61-66}$ was applied, where T_0 is a reference temperature of 1273 K used here in Figure 2b. Figure 2c shows the volume thermal expansion coefficient for liquid $Au_{55}Cu_{25}Si_{20}$ alloy estimated from the derivative of $(q_1(T_0)/$ $q_1(T)$)³ versus T (slope). Even if various power law values from 2 to 3 were used in Supporting Information Figure S1, the same conclusion can be deduced as in Figure 2b,c. It is demonstrated that the volume thermal expansion coefficient of liquid Au₅₅Cu₂₅Si₂₀ alloy is about 3.7×10^{-5} K⁻¹ below 770 K during heating and below 700 K during cooling, and changes to about $7.5 \times 10^{-5} \text{ K}^{-1}$ above 900 K during heating and above 800 K during cooling, exhibiting a reversible " λ -anomalies" in the temperature range of 700-800 K during cooling and a relatively wide temperature range of 770-900 K during heating. Figure S2 describes the full width at half-maximum (fwhm) of the first peak of S(q) as a function of temperature during heating and cooling cycles. A deviation from the linearity in the temperature dependence of the fwhm also occurs at around 770 to 900 K during heating, while upon subsequent cooling, the heating trace until a sudden deviation at around 770 K is first followed. To further examine the " λ anomalies" in volume thermal expansion coefficients, the temperature-dependent heat flow during the entire heating and cooling cycle and specific heat capacity during heating of liquid Au₅₅Cu₂₅Si₂₀ alloy are measured in Figure 3. After the melting process at around 630 K, the heat flow of liquid Au₅₅Cu₂₅Si₂₀ alloy exhibits a broad dip at around 823 K during heating and around 740 K during cooling. The temperatures for the broad dip during heating and cooling remain almost unchanged using various heating rates of 5, 10, and 20 K/min in Figure 4. The

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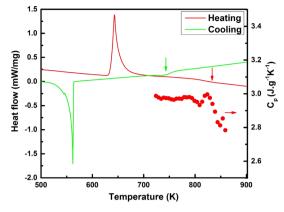


Figure 3. DSC measurements. Heat flow for $Au_{55}Cu_{28}Si_{20}$ alloy during heating (red line) from 500 to 930 K and during cooling (green line) from 930 to 500 K using a rate of 20 K/min in a high-purity Ar atmosphere by PerkinElmer DSC-7. The endothermal peak at 630 K during heating and the exothermal peak at 560 K during cooling are melting and solidification reactions, respectively. The specific heat capacity data (circles) C_p as a function of temperature for liquid $Au_{55}Cu_{25}Si_{20}$ alloy during heating from 717 to 852 K using a 5 K-step heating mode, a heating rate of 5 K/min and isothermal holding time (300 s) after each interval measured in a high-purity Ar atmosphere by PerkinElmer DSC-7.

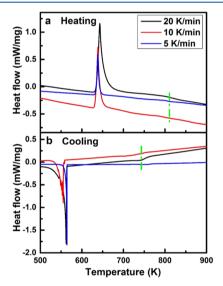


Figure 4. DSC measurements using different heating and cooling rates. Heat flow for $Au_{55}Cu_{25}Si_{20}$ alloy from 500 to 900 K using heating rates from 5, 10, and 20 K/min in a high-purity Ar atmosphere by PerkinElmer DSC-7, (a) during heating and (b) during cooling. The temperature for the broad dip marked by dash lines remains almost unchanged using various heating rates of 5, 10, and 20 K/min, indicating that the observed LLT in liquid $Au_{55}Cu_{25}Si_{20}$ alloy does not depend on the heating rate. Temperatures for the machine PerkinElmer DSC-7 were calibrated using standard pure In and Sn for different heating rates used here.

heat capacity, measured during heating, initially decreases very slowly with temperature from 723 to 800 K, but a distinct heat capacity peak emerges around 835 K. It is striking that the observed broad dip in heat flow and heat capacity peak both lie in the same temperature ranges as the transitions observed in the temperature-dependent first peak position of S(q) (Figure 2a) during heating. Small-angle X-ray scattering results for liquid $Au_{55}Cu_{25}Si_{20}$ alloy performed in a temperature range from 673 to 1174 K during the entire heating and cooling cycle

at a rate of 10 K/min in Figure 5 reveal that the data of scattering intensity versus q vector in the range of q = 0.2-5.2

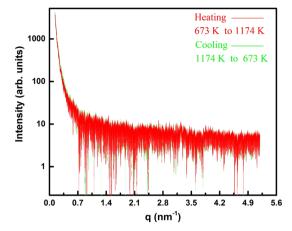


Figure 5. Small-angle X-ray scattering measurements. Small-angle X-ray scattering measurements for liquid $Au_{55}Cu_{25}Si_{20}$ alloy in a temperature range from 673 to 1174 K during the entire heating and cooling cycle using a rate of 10 K/min. No phase separation in the studied temperature range and q range of q=0.2-5.2 nm⁻¹ was observed for liquid $Au_{55}Cu_{25}Si_{20}$ alloy. During the entire heating and cooling cycle, the scattering intensity data as a function of q overlapped each other.

 nm^{-1} during heating are overlapped with those during cooling. No phase separation in the studied temperature range is observed in the liquid $Au_{55}Cu_{25}Si_{20}$ alloy. All experimental results obtained above point out that the liquid $Au_{55}Cu_{25}Si_{20}$ alloy exhibits a temperature-induced LLPT from a low-temperature liquid phase below about 700 K to a high-temperature liquid phase above about 900 K at ambient pressure, which is reversible during heating and cooling.

For pure liquids, it is known that samples solidified from liquids at different temperatures could have different structures. Thus, we further studied the Au₅₅Cu₂₅Si₂₀ alloys quenched from high-temperature (above 900 K) and lowtemperature (below 750 K) liquid phases by the melt-spinning method. Figure S3a shows their XRD patterns, demonstrating that the alloy quenched below 750 K contains some crystals, while the alloy quenched above 900 K is pure amorphous, confirming that our two samples at 300 K solidified from lowand high-temperature liquid phases indeed exhibit slight different structures. Moreover, DSC curves in Figure S3b exhibit slightly different glass transition and crystallization temperatures in both alloys and crystallization exothermal heat releases of 14.6 J/g for the alloy quenched below 750 K and of 15.5 J/g for the alloy quenched above 900 K. Despite the existence of LLPT in liquid Au₅₅Cu₂₅Si₂₀ alloy, previous studies revealed that the AuSi-based metallic liquids might have a surface crystalline layer.^{67–71} Hence, the influence of the surface crystalline layer to the LLPT should be considered. In AuSi-based metallic liquids, the surface crystalline layer has a thickness of about 6-7 atomic layers, about 2 nm. 69-71 For the liquid sample studied here with a diameter of 0.7 mm, if the surface crystalline layer would exist in our sample, the volume ratio of the surface crystalline layer to the total sample volume is roughly estimated to be about 6×10^{-6} . On the other hand, the thermal expansion coefficient of the bulk liquid $Au_{55}Cu_{25}Si_{20}$ alloy is about 3.7×10^{-5} K⁻¹ below 770 K during heating and changes to about $7.5 \times 10^{-5} \text{ K}^{-1}$ above 900 K

during heating. The change of the thermal expansion coefficient of the bulk liquid $Au_{55}Cu_{25}Si_{20}$ alloy during the transition is two times. Because the thermal expansion coefficient of bulk liquid $Au_{55}Cu_{25}Si_{20}$ alloy is strongly linked with the average atomic bond strength in the whole sample; therefore, the change in the thermal expansion coefficient of the bulk liquid $Au_{55}Cu_{25}Si_{20}$ alloy cannot be caused by minor 6 \times 10^{-6} volume fraction of the surface crystalline layer.

Figure 6a,b shows a series of integrated diffraction patterns of the $Au_{55}Cu_{25}Si_{20}$ MG during the compression up to 36.8

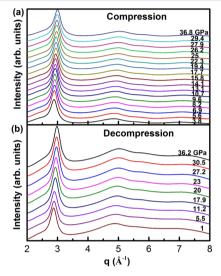


Figure 6. In situ high-energy high-pressure synchrotron XRD patterns. In situ high-pressure XRD patterns of $Au_{55}Cu_{25}Si_{20}$ MG in a DAC. (a) Integrated XRD patterns upon compression up to 36.8 GPa. (b) Integrated XRD patterns at selected pressures during decompression in the pressure range from 36.2 to 1 GPa.

GPa, followed by decompression at ambient temperature, respectively. During the entire compression and decompression range, the sample retains a fully amorphous structure, as examined from the smooth broad patterns in Figure 6, suggesting no pressure-induced crystallization reaction in the studied pressure range. The inverse first and second diffraction peak positions, $2\pi/q_1$ and $2\pi/q_2$, are shown in Figure 7a, which

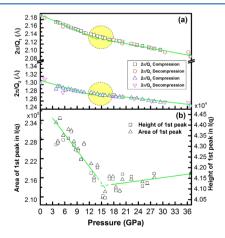


Figure 7. Structural changes. (a) Inverse peak positions $2\pi/q_1$ and $2\pi/q_2$ of $Au_{55}Cu_{25}Si_{20}$ MG upon compression and decompression as a function of pressure. (b) Pressure-induced changes of area (or intensity) and height of the first peak in I(q).

are deduced from the diffraction peak fitting. A fitting procedure is described in Supporting Information. The $2\pi/q_1$ and $2\pi/q_2$ exhibit a monotonic decrease with an obvious deviation from the linearity to the fitted curves around 13 GPa during compression. The deviation proceeds further and then agrees linearly again with the fitted curves of $2\pi/q_1$ and $2\pi/q_2$ as the function of pressure above 15.8 GPa. During decompression, both $(2\pi/q_1)$ and $(2\pi/q_2)$ are reversible. Pressure-dependent peak height and their respective area (or intensity) of the first peak in I(q) are plotted in Figure 7b. It is clear that peak height and intensity decrease with pressure in the range of 0-15 GPa, and then above about 15 GPa, both increase with pressure. The abnormal pressure-dependent $2\pi/$ q_1 , $2\pi/q_2$, and the first peak height and area all demonstrate an existence of possible two glassy phases separated by a pressure region of 13.1 to 15.8 GPa during compression, revealing a pressure-induced reversible AAPT in the Au₅₅Cu₂₅Si₂₀ MG.

Next, we address structural differences between low-temperature and high-temperature phases in the liquid $Au_{55}Cu_{25}Si_{20}$ alloy at ambient pressure and low-pressure and high-pressure amorphous phases in the $Au_{55}Cu_{25}Si_{20}$ MG at ambient temperature by AIMD simulations. Figure 8

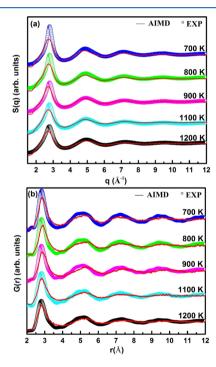


Figure 8. S(q) and G(r) for liquid $Au_{55}Cu_{25}Si_{20}$ alloy. (a) Structure factors S(q) and (b) reduced pair distribution functions G(r) for liquid $Au_{55}Cu_{25}Si_{20}$ alloy at five selected temperatures acquired from synchrotron radiation XRD measurements (open circles) and AIMD simulations (solid lines).

demonstrates structure factors S(q) and reduced pair distribution functions G(r) of liquid $Au_{55}Cu_{25}Si_{20}$ alloy at five selected temperatures acquired from synchrotron radiation XRD measurements and AIMD simulations. The consistency, in terms of peak positions, amplitudes, and shapes, between the data extracted from AIMD simulations and XRD measurements is ascertained, indicating that AIMD simulations capture the major feature of atomic configurations in the liquid $Au_{55}Cu_{25}Si_{20}$ alloy. From Au-, Cu-, and Si-centered partial pair correlation functions (PPCFs), g(r), at various temperatures in

Supporting Information Figure S5, it is found that the peak positions and the coordination numbers (CNs) of the first shell in Au- and Cu-centered PPCFs in Figure 9a decrease with

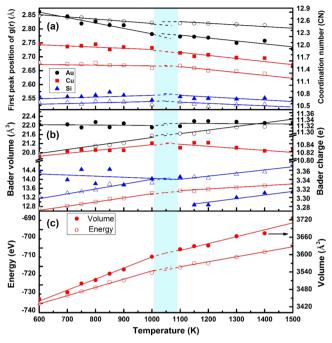


Figure 9. Temperature-induced structural evolution via AIMD simulations. Temperature dependent structural evolution for liquid $Au_{55}Cu_{25}Si_{20}$ alloy obtained from AIMD simulations. Filled symbols and open symbols are for right and left y axes, respectively, while solid lines are a guide to the eye only. (a) Peak positions and CNs of the first peak in Au- (circles), Cu- (squares), and Si-centered (triangles) g(r) in liquid $Au_{55}Cu_{25}Si_{20}$ alloy obtained from AIMD simulations. (b) Bader volumes and Bader charges of Au (circles), Cu (squares), and Si (triangles) atoms in liquid $Au_{55}Cu_{25}Si_{20}$ alloy obtained from AIMD simulations. (c) Total energy (open circles) and volume (filled circles) of the liquid $Au_{55}Cu_{25}Si_{20}$ system used in AIMD simulations.

rising temperature. Kinks are observed at around 1000-1100 K, which reveal that temperature dependences of the peak positions and the CNs of the first shell in Au- and Cu-centered PPCFs at the low-temperature phase differ from those at the high-temperature phase of liquid Au₅₅Cu₂₅Si₂₀ alloy. For the Sicentered first shell, both the temperature-dependent parameters at the low-temperature phase deviate at around 1000-1100 K from those at the high-temperature phase, which is supported by the Bader charge and volume analyses⁷² as shown in Figure 9b, in which atomic volume is uniquely defined by the charge density topology as the region bound by zero flux surfaces and the charge within the Bader volume. During cooling, Bader charge of Si atoms in the liquid Au₅₅Cu₂₅Si₂₀ alloy suddenly increases at 1100 K by a value of 0.6e per atom, while they slightly decrease for Au and Cu atoms. Besides, kinks are also detected for the temperaturedependent Bader volumes of Au, Cu, and Si atoms at around 1000-1100 K as shown in Figure 9b and the temperaturedependent total energy and volume of the system as shown in Figure 9c.

Figure S6 in Supporting Information shows total pair correlation function and PPCF g(r) at selected pressures obtained by AIMD simulations at ambient temperature. $g_{Au-Au}(r)$ is the major among all partial PCFs, followed by $g_{Au-Cu}(r)$ and then $g_{Au-Si}(r)$, contributing to total PCF, while other bonds, for example, Cu-Cu, Cu-Si, and Si-Si bonds, are almost negligible. It is found that Au-Cu and Au-Si bonds are to a greater extent densely packed as compared to Au-Au bonds. The Pressure-dependent first peak position of g(r) and volume of the system are plotted in Figure 10a, while pressuredependent energy and density of the system are shown in Figure 10b. All data reveal different pressure dependences of all four parameters below and above around 13 to 15.4 GPa. Two different glassy phases revealed in experiments in Figures 6 and 7 are supported by these AIMD results, that is, in the lowpressure region below ~13 GPa, the system energy increased relatively slow and the system density increased relatively fast with pressure as compared to these in the high-pressure region above 15.4 GPa. By using Voronoi tessellation analysis, the

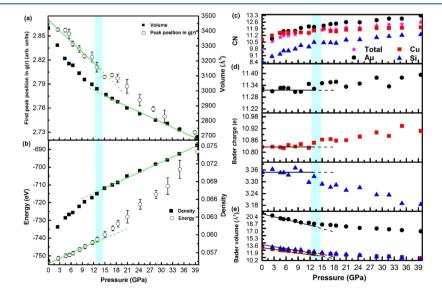


Figure 10. Pressure-induced structural evolution via AIMD simulations. (a) First peak position in g(r) and volume of $Au_{55}Cu_{25}Si_{20}$ MG as a function of pressure. (b) Pressure-dependent energy and density of $Au_{55}Cu_{25}Si_{20}$ MG. The pressure dependences of CNs (c), Bader charge (d), and Bader volume (e) of Au, Cu, and Si atoms in $Au_{55}Cu_{25}Si_{20}$ MG.

evolution of atomic packing is analyzed. The details of this technique can be found in Supporting Information. The top eleven major VP fractions for Au-, Cu-, and Si-centered atoms under various pressures are plotted in Figure S7. The most abundant VPs centered by Au atoms are CN = 11, (0,2,8,1), CN = 12, $\langle 0,3,6,3 \rangle$ and $\langle 0,2,8,2 \rangle$ in the low-pressure phase, and they are shifted to VPs CN = 13, $\langle 0,3,6,4 \rangle$ and $\langle 0,1,10 2 \rangle$ in the high-pressure phase. Silicon atoms favor low-coordinated polyhedra at the low-pressure phase, that is, CN = 8 (0.4,4.0), CN = 9 (0.3.6.0) and CN = 10 (0.3.6.1), (0.2.8.0), while above the transition pressure, their fractions are significantly reduced and they shift to higher CN = 11 (0,2,8,1) and CN = 12(0,3,6,3) VPs. The average total and partial CNs of Au-, Cu-, and Si-centered atoms at various pressures in Au₅₅Cu₂₅Si₂₀ MG are plotted in Figure 10c. At zero pressure, they are found to be $CN_{total} = 9.7 \pm 0.1$, $CN_{Au} = 10.1 \pm 0.1$, $CN_{Cu} = 10.8 \pm 0.1$, and CN_{Si} = 8.5 \pm 0.2, while at 39.6 GPa, CN_{total} = 12.5 \pm 0.1, $CN_{Au} = 13.2 \pm 0.1$, $CN_{Cu} = 12.1 \pm 0.1$, and $CN_{Si} = 11.3 \pm 0.2$. At low-pressure below about 10 GPa, the CN_{total}, CN_{Au}, CN_{Cu}, and CN_{Si} increase with pressure, while at higher pressure above about 10-15 GPa, their rates of increment become much slow. The Bader charge and volume analyses of Au, Cu, and Si elements in Au₅₅Cu₂₅Si₂₀ MG (Figure 10d,e) reveal that in the low-pressure range below about 12 GPa, no obvious changes are observed in the charge transfer among elements as shown by the guided line within uncertainty. Bader charge of Si atoms in Au₅₅Cu₂₅Si₂₀ MG starts to decrease at around 13 GPa, while they slightly increase for Au and Cu atoms. Kinks are detected for pressure-dependent Bader volumes at around 11.7 to 15 GPa with different slopes of Au, Cu, and Si atoms. The size of Si atoms is larger than that of Cu atoms at zero pressure, and it decreases relatively fast with pressure and becomes smaller than Cu atoms at 39.6 GPa, which is due to the charge transfer from Si to Cu and Au atoms. The sharp shift of low coordinated polyhedra to high coordinated polyhedral, the total and partial CN changes around Au- and Si-centered atoms, and the unusual behavior of charge transfer, the phenomena which were observed from AIMD simulations, support the experimental observation in Figures 6 and 7, that is, the existence of pressure-induced AAPT in the Au₅₅Cu₂₅Si₂₀ MG at around 10-15 GPa and ambient temperature, from both atomic packing structure and electronic structure points of view.

4. CONCLUSIONS

In summary, polyamorphic transitions of LLPT and AAPT in the Au₅₅Cu₂₅Si₂₀ alloy system have been investigated by both experimental XRD and DSC measurements and ab initio molecular dynamics (AIMD) simulations. We found solid experimental evidence for the existence of both temperatureinduced LLPT in the liquid Au₅₅Cu₂₅Si₂₀ alloy in a temperature range of 700-900 K at ambient pressure and pressure-induced AAPT in the $Au_{55}Cu_{25}Si_{20}$ MG from ~13.1 to ~15.8 GPa at ambient temperature. The temperatureinduced LLPT is characterized by "reversible λ -anomalies" of the volume thermal expansion coefficient, heat flow, and heat capacity peaks at the transition, together with different temperature-dependent first peak positions of Au-, Cu-, and Si-centered PPCF, CN of the Au-, Cu-, and Si-centered first shell, Bader volume and charge of Au, Cu, and Si atoms, and total energy and volume of the system below and above the transition. The pressure-induced AAPT is characterized by different pressure dependences of peak position, volume,

energy, density, CN, and Bader volumes and charges of the Au-, Cu-, and Si-centered atoms in the system below and above the AAPT. These changes in the system suggest that the temperature and pressure lead to the collapse of initial atomic configuration of the local structure in the Au₅₅Cu₂₅Si₂₀ alloy, and their effects are not restricted only by the changes in the atomic arrangement but also in the electronic structure. Most likely, the temperature- and pressure-induced electronic transfer from Si atoms to Au and/or Cu atoms play an important role for both temperature-induced LLPT and pressure-induced AAPT in the $\bar{Au}_{55}Cu_{25}Si_{20}$ alloy. This unique finding, that is, the existence of both temperature-induced LLPT and pressure-induced AAPT in the Au₅₅Cu₂₅Si₂₀ alloy, will trigger more challenging studies in disordered systems, especially metallic systems. For example, (1) what kind of systems could have both LLPT and AAPT? (2) What are the exact atomic structures from a microscopic point of view in both low-temperature or low-pressure, and high-temperature or high-pressure phases in the Au₅₅Cu₂₅Si₂₀ alloy? (3) Is there any kind of hidden correlation of the LLPT with the AAPT in the Au₅₅Cu₂₅Si₂₀ alloy? (4) What are the order parameters for the LLPT and AAPT in disordered materials in general? (5) Are they only the strong-to-fragile transition or similar to the polymorphism in crystalline materials?

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jpcc.9b05167.

Temperature-dependent parameter of $(q_1(T_o)/q_1(T))^n$ and the slopes of $(q_1(T_0)/q_1(T))^n$ vs T for liquid $Au_{55}Cu_{25}Si_{20}$ alloy; fwhm of the first peak of S(q) versus temperature during thermal cycles; XRD and DSC scattering measurements for Au₅₅Cu₂₅Si₂₀ alloys quenched from two different liquid states; fitting of the first peak of scattering intensity; Au-, Cu-, and Sicentered, Au-Au, Cu-Cu, and Si-Si PPCFs, g(r), obtained from AIMD simulations at selected temperatures for liquid Au₅₅Cu₂₅Si₂₀ alloy; the total and PPCFs, g(r) of Au₅₅Cu₂₅Si₂₀ MG at various pressure obtained via AIMD simulation; Voronoi tessellation analysis to analyze the evolution of atomic packing; structural features obtained from Voronoi tessellation; and the distribution of the most abundant Voronoi polyhedra (a) total (b) Au-, (c) Cu- and (d) Si-centered atoms in Au₅₅Cu₂₅Si₂₀ MG during compression (PDF)

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J.-Z.J. designed and supervised this study, S.A.K. prepared the samples and AIMD simulations, A.S.A., X.-D.W., Q.-P.C., D.-X.Z., Y.-Z.F., and J.-Z.J. conducted measurements, S.A.K., X.-D.W., Q.-P.C., D.-X.Z., Y.-Z.F., H.W., and J.-Z.J. contributed to data analyses and discussion, and J.-Z.J. wrote the manuscript.

Notes

The authors declare no competing financial interest.

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