Surface Phases and Surface Freezing in an Ionic Liquid

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Abstract

Room temperature ionic liquids (RTILs), a novel class of liquid salts, are intensively studied for their basic science and numerous emerging applications. When undercooled, RTILs comprising long alkyl chains often exhibit liquid crystal (LC) bulk phases. However, only one molecular-resolution experimental structure study was published for their LC surface phases. We measured the temperature evolution of another LC surface phase, using surface specific Å-resolution x-ray methods. This phase's existence range, 90°C, much exceeds the corresponding bulk phase's 3°C. Its thickness, L, confirms the theory-predicted logarithmic temperature dependence, with an amplitude equalling the bulk correlation length. Surprisingly, at L's divergence temperature, a ~ 20 Å-thick, hexagonally-packed, crystalline monolayer forms at, and fully covers, the sample's surface. It is identified as a surface-frozen Langmuir-Gibbs film, and fundamentally differs from the only reported RTIL surface crystal, a Coulomb-dominated, four-layer, island phase, covering only 5%-15% of the surface.

Introduction

RTIL are molten salts, comprising an organic cation with one or more alkyl chains, and a (usually smaller) organic or inorganic anion. The ions' bulky and irregular shape inhibits their solidification down to room temperatures. Short-chain RTILs, having relatively high ion mobilities, have been intensively studied for electrochemical applications in batteries, supercapacitors, solar cells, etc. ¹⁻⁴ Tribological RTIL studies ⁵ reveal their electrotunable lubricity ⁶ and quantized friction in nano-thin films. ⁷ For long-chain RTILs, like those studied here, the recently-emerged and explosively-growing field of surface active ionic liquids (SAILs), employs the long-chain-cation's amphiphilicity in medical and pharmaceutical applications ^{4,8,9}, e.g. for suppressing and dissolving Alzheimer- and Parkinson-causing amyloids ¹⁰, in micellar catalysis and synthesis ¹¹, in enhanced oil recovery (EOR) ¹², and more. In all these applications, and in particular in SAILs, surfaces and interfaces play an important role, rendering the elucidation of the RTILs' molecular-level organization and structure at interfaces, the subject of the present study, of prime importance and broad impact.

RTILs and their surfaces are of great importance and interest for fundamental science as well. Unlike Coulomb-dominated molten inorganic salts, RTILs comprise a near unique array of intermolecular interactions, including van der Waals (vdW), Coulomb, $\pi-\pi$, dipole, solvophobic, and hydrogen bonding. ^{13,14} Also unlike molten inorganic salts, the RTIL's molecular structure allows tuning the relative strengths of the two main interactions, Coulomb and vdW, by varying the cation's alkyl chain length, n. ^{15–17} Typically, for $n \le 4-6$ the dominant bulk interaction is mostly Coulomb. Increasing n, renders vdW increasingly stronger, driving the bulk from a locally isotropic, homogeneous and unstructured liquid, to a locally structured one, where apolar chains segregate from the charged moieties into domains correlated over n-increasing nanoscale lengths. ^{15,16} Still-longer chains induce a bulk LC phase, with the n-threshold for the LC phase in a given homologous cation series depending on the anion. ^{18–20} While this bulk phase sequence has been well studied experimentally (calorimetry, x-ray diffraction, etc.) for several RTIL series ^{14,21,22}, that of the RTIL/air surface has

been much less studied with sub-molecular resolution $^{23-28}$, since the required surface specific x-ray measurements are rather challenging, and require special setups. $^{29-31}$ In fact, only a single x-ray study of a LC surface phase of an RTIL, $[C_{22}mim][NTf_2]$, has been published to date, identifying the phase as smectic A, and determining the temperature dependence of the thickness of the LC surface layer. 32

Here we report an x-ray study of the liquid/air surface of another long-chain RTIL, [C₂₀mim][NTf₂], and its evolution with temperature upon cooling from a conventional surface-layering phase^{27,28}, through the emergence and growth of a LC surface phase, to the eventual divergence of this LC surface layer's thickness to engulf the sample's bulk. Moreover, unlike [C₂₂mim][NTf₂], a surface-frozen (SF) crystalline monolayer is found to appear at the RTIL/air surface upon formation of the bulk LC phase. Crystalline surface order in an RTIL has been hitherto reported only for a single Coulomb-dominated short-chain compound³³, where crystalline "islands" were found to coexist at the surface with the surrounding liquid surface "sea". The islands consisted of stacks of four monomolecular layers having a rectangular lateral molecular packing. The crystalline surface monolayer found here fully covers the surface, is vdW-dominated, hexagonally-ordered, and closely resembles, albeit with distinct differences, the SF monolayer on alkane melts.³⁴ The details and temperature dependence of the LC surface phase, and those of the SF monolayer, are presented and the underlying molecular-scale physics is discussed.

Experiment

Materials

The samples studied were 1-alkyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imides, (denoted $[C_n mim]^+[NTf_2]^-$), where n = 18, 20, purchased as powders at room temperature from Iolitec, Germany, with manufacturer-stated purities > 98% (n = 18) and > 95% (n = 20). For further details see Supplementary Information (SI).

x-ray methods

Since the x-ray methods used are well-documented^{34–36}, we present here only a brief summary. Further details are given in the SI.

X-ray reflectivity: The laterally-averaged surface-normal electron density profile, $\rho(z)$, where z is the surface-normal coordinate pointing into the sample, is obtained from the measured x-ray reflectivity (XRR). $^{34-36}$ XRR is the surface-reflected intensity fraction $R(q_z)$ of an x-ray beam impinging on the free surface of the sample at a grazing angle α . $q_z = (4\pi/\lambda)\sin(\alpha)$ is the surface-normal scattering vector and λ - the x-rays' wavelength. Using physical considerations, a mathematical model of $\rho(z)$ is constructed and its calculated XRR curve is least-squares computer-fitted to $R(q_z)$ to yield the model-defining parameters. $^{34-36}$ Here we employ the model of Ref. 32 (see details in SI), depicted schematically in Fig. 1: coming in along the z axis from the RTIL/air free surface at z=0, this model comprises first a layered smectic LC surface phase, extending from z=0 to z=L (Fig. 1(a,b)). From z=L down lies a "normal" layered phase (Fig. 1(a,d)), with the layering decaying exponentially with z, as we found for non-LC, shorter-chain, RTILs. 27,28 Both phases were assumed to have a common layer spacing d. 32

Grazing incidence diffraction and Bragg rods The lateral order within the sample's surface layer was explored by Grazing Incidence Diffraction (GID). $^{34-36}$ Here a vertically-aligned linear detector is rotated horizontally by an angle θ away from the x-ray reflection plane $(\theta = 0^{\circ})$. Each pixel along the detector corresponds to scattering vector components $q_z = (2\pi/\lambda)(\sin \alpha + \sin \beta)$ and $q_x = (2\pi/\lambda)\sin \theta \cos \beta$, where β is the vertical angle from the sample plane of an x-ray scattered into the pixel, and $\lambda = 0.5586$ Å. Thus, a GID scan produces a scattered intensity map in the (q_x, q_z) plane. The map yields two conventional plots: the GID and the Bragg rod (BR) plots. The first is the q_z -integrated intensity vs. q_x , showing the diffraction peaks of the lateral order in the surface layer. The second is

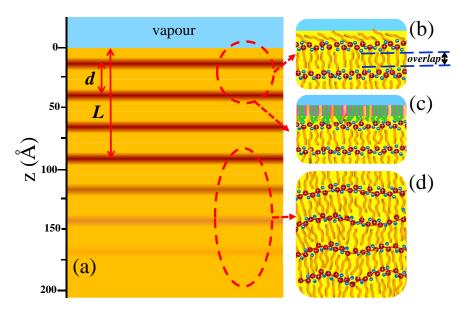


Figure 1: (a) Cartoon depicting the surface normal structure of $[C_{20} \text{mim}][\text{NTf}_2]$. Narrow brown, and broad orange, stripes represent, respectively, the charged and alkyl moieties' layers. The LC phase extends down to z=L, with negligible decay of the layering order. Below it lies the normal layering phase, decaying much faster with depth. d is the layer spacing. (b) Near-surface layers' details at $T>53^{\circ}\text{C}$, also showing the air-protruding alkyl chains of half the top-layer cations, and the chain overlap within the alkyl layer. (c) Near-surface layers' details at $T=53^{\circ}\text{C}$, showing the surface-frozen crystalline monolayer (purple slab), comprising the mixed, stretched, vertical, upper 20Å of the air-protruding cation tails (yellow) and of the alkyl-bromide guest molecules (green). (d) The z-decay of the "normal" layering order. The decay starts at the surface (z=0) at temperatures where no LC surface phase exists, or below the LC phase where it does. The various phases and their structures are discussed in the Results and Discussion sections.

the intensity distribution along q_z at the q_x position of a peak in the GID plot, yielding information on the ordered layer thickness and the molecular tilt.^{34–36}

Using for the GID scans $\alpha = 0.04^{\circ}$, $\sim 30\%$ below the critical angle for total external reflection $\alpha_c = 0.056^{\circ}$, the surface-exploring beam is evanescent, its intensity decaying exponentially with depth with a ~ 65 Å decay length. Thus, it samples only the near-surface region of the RTIL²⁸, and avoids swamping the weak surface-signal by the strong bulk-signal present for macroscopic x-ray penetration when $\alpha > \alpha_c$.

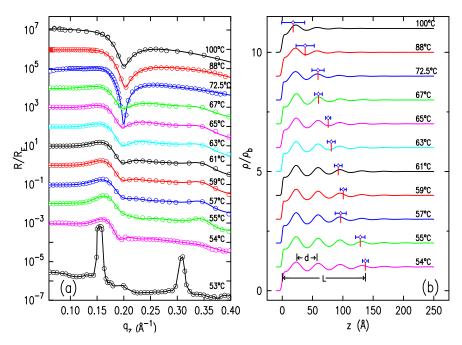


Figure 2: (a) Measured (symbols) and model fitted (lines) Fresnel-normalized x-ray reflectivity curves of $[C_{20}mim][NTf_2]$ at the temperatures listed. The line for 53°C is a guide to the eye only. (b) Bulk-density-normalized surface-normal electron density profiles, obtained from the model fits in (a). The red vertical lines (with their blue error bars) mark the fit-derived border L between the liquid-crystal surface phase and the underlying "normal" layered RTIL phase. The layering periodicity d is also marked.

Results

Fig. 2(a) presents the temperature (T) evolution of the measured $R(q_z)$, normalized by the Fresnel XRR of an ideally smooth and abrupt surface³⁵, $R_F(q_z)$, (symbols), and their model fits (lines), for $[C_{20}\text{mim}][\text{NTf}_2]$. Fig. 2(b) shows the fit-derived $\rho(z)/\rho_b$, where ρ_b is the bulk electron density, and the LC layer thickness L. The $T \geq 72.5^{\circ}\text{C}$ curves in (a) closely resemble those of $n \leq 18$ in Ref. 28, which have no LC surface phase. The single negative peak observed for $T \geq 72.5^{\circ}\text{C}$ at $q_z \sim 0.18 \text{ Å}^{-1}$ corresponds to $d \sim 35 \text{ Å}$, in agreement with our previous studies, where the present 100 °C curve already appeared. The very thin LC surface layer at these high T's is insufficient to significantly impact the shape of R/R_F . Cooling from $T = 100^{\circ}\text{C}$ extends the "normal" layering deeper, as expected, rendering the negative peak sharper and larger. Below $T \leq 67^{\circ}\text{C}$, phase shifts in the interfering components of the x-rays reflected from the temperature-varying stacked LC and underlying "normal" layers

wash out the negative peak. The increasing number of layers within the LC surface phase, reflected in the increasing L, gradually intensifies the two orders of a LC quasi-peak, at $q_z \approx 0.17$ and 0.34 Å^{-1} . Eventually, at $T \approx 53 \text{ °C}$ the LC surface layer's thickness diverges to engulf the bulk, yielding the sharp, intense, two orders of the LC peak observed in the bottom curve of Fig. 2(a).

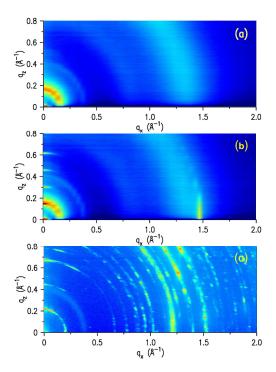


Figure 3: (a) Measured GID intensity map at T=54 °C, for $\alpha<\alpha_c$. The typical nanostructured RTIL liquid rings at $q_x^I\approx 0.18$ Å⁻¹, $q_x^{II}\approx 0.85$ Å⁻¹ and $q_x^{III}\approx 1.38$ Å⁻¹, are clearly visible. (b) The same map at T=53 °C, at the onset of the bulk LC phase. Additional to the liquid rings, four diffraction orders of the thick LC phase are observed along q_z . A Bragg rod is observed at $q_x\approx 1.49$ Å⁻¹, indicative of an hexagonally packed, SF crystalline monolayer of extended vertically-aligned molecules.(c) The same map for [C₁₆mim][NTf₂], where no LC phases exist either at the bulk or the surface. This map was measured at T=37.5 °C, the liquid-to-crystal transition. For a discussion see text.

To search for possible lateral molecular ordering at the sample/air surface we have carried out GID measurements at the same temperatures where XRR curves were measured, using $\alpha \approx 0.7\alpha_c$ as detailed in the Experimental section. The GID intensity map in Fig. 3(a), measured at T = 54°C, shows the three main liquid RTIL rings, reported and discussed for [C₂₀mim][NTf₂], and its homologous series, in our recent bulk and surface studies. ^{15,28} The

strong "low-q" ring I at $q_x^I \approx 0.18$ Å⁻¹, is due to nanoscale segregation of apolar cationic alkyl tails from the charged moieties, causing the bulk nanostructuring ^{16,17}, and also the conventional surface layering. ^{27,28} Rings II and III at $q_x^{II} \approx 0.85$ Å⁻¹ and $q_x^{III} \approx 1.38$ Å⁻¹ are due to correlations of same-charge ions, i.e. the cationic headgroups and anions (II), and of intra- and inter- molecular adjacent atoms, mostly the adjacent alkyl chains (III). ^{14,16,17} Detailed simulations support these assignments. ^{37–39} It is important to note that in the Fig. 3(a) map no sharp structural diffraction peaks, assignable to long-range lateral order, are observed, and we conclude that no such order exists. This holds true at all measured temperatures $T > 53^{\circ}$ C.

At T=53°C (Fig. 3(b)), where the LC surface phase diverges to engulf the bulk as explained above, two new features appear, which we now discuss in some detail. The first feature is a strong, saturated, peak at 0.16-0.17 Å⁻¹ and its three higher-order harmonics, appearing as short horizontal "dashes" along the q_z -axis, also observed in other lamellar LC phases. ⁴⁰ The two lowest-order "dashes" correspond to the peaks appearing in Fig. 2(a) at T=53 °C. Since the LC phase here is smectic-A ³², the appearance of an intense second-order peak is surprising, in view of its absence in conventional, non-RTIL, thermotropic smectic-A LCs. ^{41–43} This must be due to the scatterers' form factor being non-centrosymmetric, in contrast with the centrosymmetric form-factor in conventional, non-RTIL, smectic-A LCs. ⁴³

The second, and most surprising, feature observed in Fig. 3(b) is a short, thin, and intense diffraction line at $q_x^{BR} \approx 1.5 \text{ Å}^{-1}$. The shape, orientation, and extension of this feature mark it as a Bragg Rod^{34,35}, the signature of a crystalline surface monolayer of extended, surface-normal-oriented, molecules. Only a single case of a surface crystal has been hitherto reported for RTILs. However, this occurred for a short-chain, Coulomb-dominated RTIL molecule³³, where the surface crystals were rectangularly-packed, 4-layered, "islands" covering only 5%-15% of the surface area, in coexistance with the liquid surface "sea". In contrast, the single BR observed here indicates a hexagonal packing ^{34,44} and its absolute intensity and persistence with unchanging intensity in repeated measurements at different

surface positions, imply that it fully covers the surface of the sample.

Finally, the fact that the LC harmonic peaks along q_z in Fig. 3(b) form no rings but stay closely confined to the q_z axis indicates that the LC phase, even though now macroscopically thick, preserves the surface-parallel orientation of the smectic layers, a property important for tribological applications. ^{6,45} When RTILs freeze into a fully crystalline phase the surface and underlying bulk break up into randomly oriented crystallites, thus forming rings in the diffraction pattern. An example is shown in Fig. 3(c) for [C₁₆mim][NTf₂], where the RTIL's surface and bulk freeze from a nanostructured liquid into a crystalline phase, without an intermediate LC phase.

We now proceed to discuss these experimental results in the section.

Discussion

The surface-normal structure

We discuss first the XRR results and their fits in Fig. 2 to elucidate the temperature evolution of the top LC surface phase and of the "normal" layering phase below it (see Fig. 1(a)). The LC layer thickness L, normalized by the layer period d, both derived from the fits, are plotted (\Box) in Fig. 4(a). Surface energy balance considerations 32,46,47 predict that $L(T)/d = (L_0/d) \ln(t_1/t)$, where $t_1 = (T_1 - T_0)/T_0$ and $t = (T - T_0)/T_0$, and T_1 and T_0 are the onset and the divergence temperatures of the LC surface layer, respectively. The good agreement of this expression (black dash line) with the fit-derived L/d values confirms the expected logarithmic T-dependence, and yields $T_0 = 51.8$ °C and $T_1 = 141.8$ °C, and thus a large $T_1 - T_0 = 90$ °C existence range for the surface LC phase. The fit also yields $L_0/d = 1.05 \pm 0.03$, which agrees well with our published d-normalized bulk correlation length 15 , $\xi_b/d = 1.06 \pm 0.01$. This close agreement confirms theoretical predictions 46 , and demonstrates that the same main

correlations dominates both bulk and surface phases. ⁴⁶ The L/d values of $[C_{22}mim][NTf_2]^{32}$ in Fig. 4(a) (\bigcirc , and fitted red dash line) are close to ours, and yield the same ~ 90 °C T-range and the same agreement of L_0/d with ξ_b/d^{32} , albeit with a larger $L_0/d = 1.2$ and $\xi_b/d = 1.25 - 1.3^{15,22}$. The smaller value of L_0/d of n = 20 as compared to L_0/d of n = 22 reflects the eduction in the vdW interaction strength with decreasing chain length, also found for the decay lengths of conventional surface layering ²⁸ and the correlation lengths of the bulk, ξ_b/d . ¹⁵

While this agreement of the results for the two close-n RTILs is not surprising, the presence of a LC surface phase in [C₂₀mim][NTf₂] has not been hitherto reported, and implies the probable existence of a corresponding LC bulk phase. Indeed, a slow, 0.5K/min, DSC scan, Fig. 4(b), reveals the existence of this phase, albeit over a short temperature range, $\sim 3^{\circ}\mathrm{C}$ only, and at a strong undercooling: $\sim 10^{\circ}\mathrm{C}$. The corresponding ranges for $[C_{22}mim][NTf_2]$ are $\sim 10^{\circ}C$ of existence and $\sim 1^{\circ}C$ of undercooling.³² It seems therefore that $[C_{20}mim][NTf_2]$ is at the lowest-n limit which induces LC phases in this homologous series. We checked this hypothesis by DSC and XRR measurements on the shorter-chain homolog [C₁₈mim][NTf₂]. Indeed, neither bulk nor surface LC phases were detected by both techniques. The large increase in undercooling, and decrease in the existence range, incurred upon reducing the cation's chain length, $n \approx 20$, by 2 methylenes only, implies that a rather small, 10%, reduction in the vdW interaction strength suffices to offset the interaction balance dominating the RTIL's bulk structure and eliminate the LC phase. Further insight into the n and T phase boundaries of the LC surface and bulk phases, and a more accurate determination of their values, may be obtained from concentration-dependent measurements on [C₁₈mim][NTf₂]/[C₂₀mim][NTf₂] mixtures. For the bulk, such binary RTIL mixtures were shown to behave as a single homolog having a concentration-interpolated non-integer n value, intermediate between those of the two components. 48 Such measurements may also provide insight into the ways that the presence of a surface modifies the bulk's non-integer n.

Note also that while the good fits obtained in Fig. 2(a) with a model using the same d

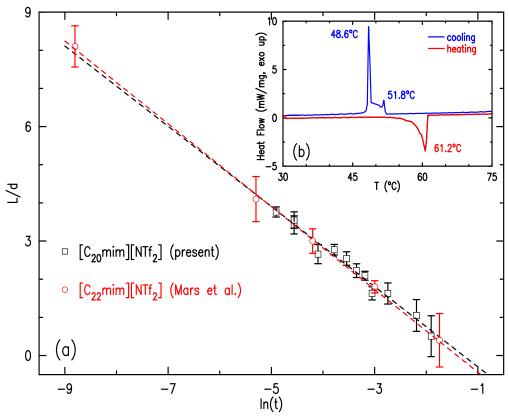


Figure 4: (a) Measured (symbols) and theory-fitted (lines) liquid-crystal surface layer thickness, L, normalized to the layering periodicity d of the specified RTILs. Both compounds conform to the predicted logarithmic T-dependence, with a slight slope difference, indicating a corresponding difference in the bulk correlation length. For discussion see text. (b) Differential Scanning Calorimetry (DSC) scan (0.5 K/min) for [C₂₀mim][NTf₂], showing melting at 61.2 °C, the formation of a strongly-undercooled bulk liquid-crystal phase at 51.8 °C and the bulk freezing at 48.6 °C.

for the microscopic few-monolayer-thick LC surface layer and for the underlying "normal" layered phase yield $d \approx 35.5$ Å, the q_z positions of the "dashes" along q_z in Fig. 3(b), and the T = 53 °C peaks in Fig. 2(a), yield for the post-divergence LC phase a larger $d_{LC} = 2\pi/q_z = 39$ Å. Since the near-surface structure in the "normal" liquid phase consists of alternating thin (~ 5 Å) layers of the charged moieties (cation headgroups and anions) and thick (~ 30 Å) apolar layers of alkyl chains ^{27,28}, the ~ 3.5 Å higher d_{LC} could result from a staggering of the headgroups in the charged moieties' layers in the macroscopic LC phase thus increasing their thickness, from an expansion of the apolar alkyl layers, or from a combination of both. A significant contribution from headgroup rearrangement is unlikely,

since such rearrangement would change the same-charge ion distances, shifting the q_x^{II} ring position. Comparing Fig. 3(b) and (a), no shift in q_x^{II} is observed. The model of flexible, kinked, interdigitated chains proposed for the alkyl layer ^{15-17,49-51} could, however, explain an alkyl layer expansion as resulting from an opposite-direction shift of adjacent interdigitated molecules along the layer-normal, decreasing the chains' overlap (Fig. 1(b)). Indeed, the bulk d was observed to be consistently larger than the surface one also for lower-n molecules in our homologous RTIL series, which do not exhibiting a LC phase, albeit with a smaller 1-2 Å difference. ¹⁵ An additional stretching of the flexible, kinked, chains by the increased lateral compression due to the stronger vdW interaction in the LC phase ³² could perhaps also account for part of this d-difference.

The surface-parallel structure

The GID measurements that probe the lateral structure of the surface show in Fig. 3(b) the appearance at $T=53^{\circ}\mathrm{C}$ of a single Bragg rod at $q_x\approx 1.5~\mathrm{Å}^{-1}$. The q_z intensity distribution of this BR, shown in Fig 5(a), peaks at $q_z\approx 0~\mathrm{Å}^{-1}$ and is $\sim 0.16~\mathrm{Å}^{-1}$ wide (HWHM) indicating, respectively, a monolayer of surface-normal scatterers $\sim 20~\mathrm{Å}$ long. The BR's integrated intensity, and its high-resolution (q_x,q_z) intensity map, are plotted in Fig. 5(b) and (c), respectively. A fit by the theoretical profile 34,35 (see SI) in Fig. 5(a) (line) yields a scatterer length of $\ell=19\pm1~\mathrm{Å}$, much shorter than either the layer spacing d or the calculated extended length of the cations' alkyl tails 34,52 , $l_n=1.27\times(n-1)+1.5=25.6~\mathrm{Å}$, and a tilt from the surface normal $\lesssim 2^{\circ}$. The Lorentzian fit (line) to the measured GID q_x -peak in Fig. 5(b) yields its position, $q_x^{BR}=1.493\pm0.001~\mathrm{Å}^{-1}$, and FWHM, $\Delta q_x^{BR}=0.022\pm0.001~\mathrm{Å}^{-1}$. This q_x^{BR} indicates a (hexagonal) lattice constant $a=4\pi/(\sqrt{3}\times1.493)=4.860\pm0.003~\mathrm{Å}$ and an area per molecule $A_{SF}=(\sqrt{3}/2)a^2\approx20.5~\mathrm{Å}^2$, typical of close-packed, extended, untilted, alkyl chains 44 , and very close to those of surface-frozen monolayers on

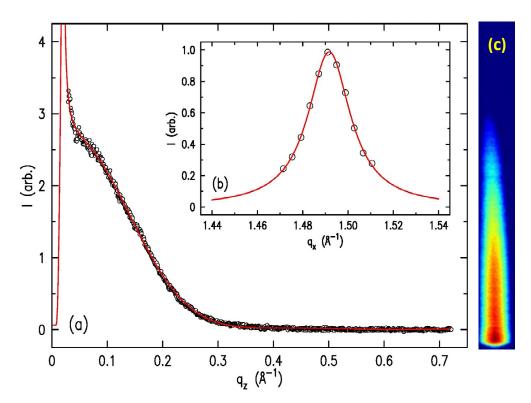


Figure 5: (a) The measured (symbols) and theory-fitted (red line) intensity distribution along the BR in Fig. 3(b). (b) The q_x -measured (symbols), and Lorentzian fitted (line), GID peak corresponding to the BR. The position and width yield, respectively, the lattice constant a=4.86 Å and crystalline coherence length $\xi_{SF}\approx 370$ Å. (c) A high resolution intensity map of the BR observed in Fig. 3(b), covering the $[q_x(\text{horizontal}), q_z(\text{vertical})]$ -region from [1.48, 0] Å⁻¹ to [1.505,0.25] Å⁻¹. See text for discussion.

alkane melts 34 and of water-supported crystalline Langmuir-Gibbs (LG) monolayers. $^{53-55}$

To elucidate the nature of our SF monolayer, we note that the molecular area of $[C_n mim][NTf_2]$ in its "normal liquid" layered surface phase²⁸, $A_{RTIL} \approx 60 \text{ Å}^2$, is threefold larger than A_{SF} above. Thus, the SF layer observed here can not consist solely of $[C_{20}mim][NTf_2]$ cationic alkyl tails, even if 100% of such tails in the topmost cation surface layer point into the vapour, instead of only 60%, as found for the "normal" liquid surface phase²⁸ and shown schematically in Fig. 1(b). Thus, we suggest that the SF monolayer is a new type of a LG film. To asses this suggestion we discuss first conventional, surfactant solution supported, LG surface films.

Conventional LG films^{53,56–58} form when a liquid alkane droplet, which does not spread on pure water, is placed on the free surface of an aqueous solution of a surfactant comprising

an alkyl tail of length similar to that of the droplet's alkane. The Gibbs-adsorbed surfactant layer residing at the solution's surface induces migration of alkane molecules from the droplet along the surface, and a monolayer of mixed alkane molecules and air-protruding surfactant tails forms on the solution's free surface. Since only a single monolayer forms, and the excess alkane remains aggregated, the mixed monolayer is clearly autophobic. ⁵⁹ The alkane molecules intrude between the air-protruding alkyl tails of the surfactants, the lateral spacing of which is determined by the surfactant's bulk concentration. Note, however, that even at maximal surfactant surface coverage, occurring at the bulk critical micelle concentration (cmc), the surfactant tails' spacing is significantly larger than that of close-packed extended alkyl chains. ⁵⁷ The intruding alkanes, vertically aligned by the surfactants' air-protruding tails, thus reduce the area/chain to the canonical $A_{alk} \approx 20$ Å² required for forming a close-packed crystalline monolayer of alkyl chains. ^{34,44} Indeed, while SF does not occur in the alkane-free Gibbs-adsorbed, pure surfactant, surface film, the mixed LG monolayer surface-freezes upon cooling, forming a hexagonally-packed crystalline monolayer of surface-normal-aligned mixed alkane molecules and surfactant tails.

We suggest, therefore, that the crystalline monolayer observed here is a surface-frozen LG film, where the air-protruding alkyl tails of the top cation layer of the sample (see Fig. 1(b)) play the role of the surface-adsorbed surfactants, and alkylated guest molecules, such as traces of the alkyl-bromide used in the synthesis, play the role of the alkanes. The structure of this crystalline monolayer is schematically depicted in Fig. 1(c). NMR spectroscopy (see SI) indeed confirms the presence of alkyl-bromide guest molecules in the sample at a ~ 2 mol% level, consistent with the material's purity and the synthetic route followed. This concentration is much more than needed for inclusion in the single SF monolayer. These molecules remain dissolved in the "normal" liquid RTIL bulk. They are squeezed out to the sample's surface only when the more-ordered bulk LC phase forms. This would explain why the SF monolayer appears only at the onset of the LC surface phase's divergence into the bulk, at 53°C.

The discussion above indicates a 3-5 fold outnumbering of the cation tails by the guest molecules in the LG monolayer. Similar and even larger, >10-fold, outnumberings were found in conventional LG films at the water/air^{53,56-58} and water/alkane⁶⁰ interfaces. Although a minority, the tail's presence in the conventional LG film was found to be mandatory for inducing SF in the monolayer.⁵⁷ This is because in both systems the surfactant-or-cation tails promote spreading^{61,62} of guest chain molecules on the surface, and orient the chains side-by-side and surface-normal, thus helping the short-range intermolecular vdW attraction to form the crystalline SF monolayer. The formation in both systems of only a single surface monolayer, implying its autophobicity⁵⁹, further supports the suggested structural similarity of the SF monolayer in both systems.

The lateral extent of the monolayer's crystalline order is characterized by the crystalline coherence length given by the Debye-Scherrer formula $^{63}\xi_{SF}=0.9\times2\pi/\sqrt{(\Delta q_x^{BR})^2-(\Delta q_x^{res})^2}\approx$ 370 Å, where $\Delta q_x^{res} = 0.016$ Å⁻¹ is the experimental resolution. This ξ_{SF} coincides with the values obtained for water-supported LG monolayers ⁵³, in line with the assignment offered here. The thinner-than-expected monolayer, ~ 20 Å only, may result from ordering of only the top parts of the chains, due to packing frustration caused by the areal mismatch between the headgroups and the chains. Similar mismatch-inhibited partial-length ordering has been observed in amorphous-SiO₂-supported monolayers of alkyltrichlorosilanes ⁶⁴ and alkanols ⁶⁵, while a nearly-matching sapphire substrate yields a full-length ordering for alkanols. ⁶⁶ Although here the alkyl tails' lower ends are not firmly anchored to solid substrate sites but rather to the cation headgroups, the headgroup size provides a hard-sphere closest approach for adjacent tails' lower ends, keeping them apart by more than the chain-preferred $\sim 4.8~\text{Å}$ as firmly as if they were substrate-anchored. Finally, while the present tentative LG-film assignment accounts for the observations, and seems therefore plausible, we can not rule out the possibility that the abrupt thickness-divergence of the LC surface layer near 53 °C somehow causes an ion rearrangement in the top surface layer such that the top 20 Å of the cationic chains are brought into close-enough contacts for their vdW interaction to induce

surface-freezing, without, or with minimal, additional guest molecules intruding between the cationic chains.

Conclusions

The present temperature-dependent x-ray and DSC study of the surface and bulk phases of $[C_{20}mim][NTf_2]$ reports for the first time the existence of liquid-crystal surface and bulk phases in this compound, and presents a number of new results, important for long-chain RTILs in general, and for their LC phases and SAILs in particular. These findings are:

- Measurement of the temperature evolution of the surface-normal structure, showing that the LC surface phases's thickness L conform to the theory-predicted 32,46 logarithmic temperature dependence, with an amplitude equalling the bulk correlation length, thus demonstrating that bulk correlations govern also the surface structure.
- A demonstration that the bulk LC phase forms by a surface-induced inhomogeneous nucleation 67,68 , through the divergence of the LC surface phase's thickness at ~ 53 °C.
- A hitherto unreported abrupt ~ 3.5 Å increase in the layer period d upon transition from a "normal" to a LC bulk phase. This shift is tentatively assigned to an opposite-direction shift of adjacent interdigitated chains within the alkyl domains, thus decreasing their overlap and increasing their headgroups' layer spacing.
- A discovery of an abrupt formation of a SF crystalline monolayer at the sample's surface at the temperature where the LC surface phase diverges to engulf the bulk.
- A detailed characterization of the crystalline monolayer's structure: vdW-dominated hexagonal packing of surface-normal alkyl chains with 20.5Å²/chain, and full surface coverage. The only other RTIL surface crystal reported to date³³ occurs for a short-chain, Coulomb-dominated RTIL, forming 4-layered, rectangularly-packed, crystalline islands covering only 5%-15% of the surface.

- Identification of the crystalline monolayer as a new type of a surface-frozen LG monolayer ⁵³, a mixture of cationic tails and alkylated guest molecules (e.g. alkyl-bromides used in the RTIL synthesis). These are squeezed out of the bulk upon formation of its LC phase, and fill the gaps between the top cation layer's air-protruding chains. All hitherto reported LG films employ aqueous surfactant solutions with surface-adsorbed surfactant monolayers. Here of course no water, or other solvent, is involved.

Similar studies on RTILs with different-size and different-shape headgroups and anions, including mixtures with alkanes and their derivatives, may provide further insight into surface freezing and surface phase behaviour in a broader range of RTILs. For example, in conventional LG films the surface density of surfactant molecules can be tuned by varying the aqueous solution's bulk surfactant concentration. In pure RTILs, however, the sizes of the anion and cation headgroups determine the surface density of cation tails. Nevertheless, this density can still be tuned by using different-size anions and cation headgroups, and cations with more than one alkyl tail. ⁶⁹

Finally, we note that currently-scarce molecular-resolution studies, similar to ours, of RTILS at solid and liquid interfaces, particularly of long-chain RTILs which are surface active, should provide fresh insights for both fundamental science, and electrochemical ^{1,3,70}, tribological ^{5–7}, and medical ^{4,8,9} applications of RTILs and SAILs.

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Supporting Information Available

Sample, experimental, and data analysis details.

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Graphical TOC Entry

