High-pressure magnetic, electronic, and structural properties of MFe_2O_4 (M = Mg, Zn, Fe) ferric spinels

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(Received 21 March 2017; revised manuscript received 27 April 2017; published 23 May 2017)

Magnetic, electronic, and structural properties of MFe₂O₄ (M = Mg,Zn,Fe) ferric spinels have been studied by ⁵⁷Fe Mössbauer spectroscopy, electrical conductivity, and powder and single-crystal x-ray diffraction (XRD) to a pressure of 120 GPa and in the 2.4-300 K temperature range. These studies reveal for all materials, at the pressure range 25–40 GPa, an irreversible first-order structural transition to the postspinel CaTi₂O₄ – type structure (Bbmm) in which the HS Fe3+ occupies two different crystallographic sites characterized by six- and eightfold coordination polyhedra, respectively. Above 40 GPa, an onset of a sluggish second-order high-to-low spin (HS-LS) transition is observed on the octahedral Fe³⁺ sites while Fe³⁺ occupying bicapped trigonal prism sites remain in the HS state. Despite an appreciable resistance decrease, corroborating with the transition to the LS state, MgFe₂O₄ and ZnFe₂O₄ remain semiconductors at this pressure range. However, in the case of Fe₂O₄, the second-order HS-LS transition on the Fe³⁺ octahedral sites corroborates with a clear trend to a gap closure and formation of a semimetal state above 50 GPa. Above 65 GPa, another structural phase transition is observed in Fe₃O₄ to a new *Pmma* structure. This transition coincides with the onset of nonmagnetic Fe²⁺, signifying further propagation of the gradual collapse of magnetism corroborating with a sluggish metallization process. With this, half of Fe³⁺ sites remain in the HS state. Thus, this paper demonstrates that, in a material with a complex crystal structure containing transition metal cation(s) in different environments, a HS-LS transition and delocalization/metallization of the 3d electrons does not necessarily occur simultaneously and may propagate through different crystallographic sites at different degrees of compression.

DOI: 10.1103/PhysRevB.95.195150

I. INTRODUCTION

Spinels are ternary compounds with a structure isomorphic to the mineral spinel (MgAl₂O₄). The general formula AB_2X_4 consists of divalent X anions and most commonly cation valences of A^{2+} and B^{3+} (known as 2-3 spinels) or A^{4+} and B^{2+} (4-2 spinels). In oxide spinels, O^{2-} anions create a distorted cubic close-packing arrangement [1]. For 2-3 normal spinels, tetrahedral and octahedral sites are occupied by A^{2+} and B^{3+} cations, respectively. In inverse spinels, half of the B^{3+} cations occupy the tetrahedral sites, and all the A^{2+} cations occupy octahedral sites. In general, spinels can be partially inverted. Many materials can be found with the spinel structure, exhibiting a variety of electronic, magnetic, and optical properties [2-4]. A few spinels are considered gemstones, specifically transparent spinels such as MgAl₂O₄, which is a hard mineral and has a high refraction index. Spinels are also used for industrial purposes, due to their hardness and stability upon exposure to high temperatures, chemicals, and physical wear [5].

The spinel structure and its high-pressure (HP) modification, the so called postspinel, are adopted by many minerals and compounds which play an important role in modeling the properties and geochemistry of Earth's transition zone and mantle. Therefore, the behavior of spinels under nonambient conditions is of considerable geophysical importance and has been the subject of several powder x-ray diffraction (XRD) studies [6-10]. These studies performed for MgFe₂O₄, MgAl₂O₄, and ZnFe₂O₄ have shown a pressure-induced first-order structural transition to more compact postspinel structures [6-10] in the pressure range 25–45 GPa. However, there is controversy regarding the ascertainment of the postspinel structure(s), with the most commonly given candidates being CaMn₂O₄, CaTi₂O₄, and CaFe₂O₄ type structures. These three structure types are orthorhombic and have the Ca²⁺ in a bicapped trigonal prism coordination (eight surrounding O²-) and the transition metal in octahedral coordination. The main difference between the three structures, besides the difference in symmetry, is that CaFe₂O₄ contains two different crystallographic sites for Fe³⁺, whereas in the other structures, all the octahedra are equivalent.

Previous studies of an archetypal spinel and the oldest recognized natural magnetic material, magnetite (Fe²⁺Fe³⁺₂O₄), utilizing ⁵⁷Fe Mössbauer spectroscopy (MS), also revealed a pressure-induced first-order phase transition at the pressure range 25–45 GPa [11], in good agreement with XRD studies [12–14]. However, according to MS measurements [11], the

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postspinel phase exhibits, in addition to the Fe^{2+} site, two distinct Fe^{3+} sites with different hyperfine interaction parameters. These MS results appear inconsistent with the abovementioned XRD studies claiming that the postspinel Fe_3O_4 phase is of the $CaMn_2O_4$ or $CaTi_2O_4$ type structures, wherein all Fe^{3+} cations should occupy equivalent crystallographic sites. The possible incongruity between the structure modeled from XRD data and the MS data can be attributed to the lack of necessary resolution due to the intrinsic broad lines of the HP XRD powder data, which may not provide sufficient information for a reliable structure assignment. Precise single-crystal XRD could be crucial in such a case for the final structural assignment within the postspinel pressure range.

The MS and electrical resistance R(P,T) studies of magnetite also revealed another important result—the absence at the pressure range up to 120 GPa of any first-order pressureinduced electronic transformations typical for Fe³⁺, namely, Mott or high-to-low spin (HS-LS) transition. Such electronic transitions are usually observed at the pressure range of 40-60 GPa and typically coincide with a first-order structural phase transition [15,16]. In the meantime, only a partial sluggish magnetic → nonmagnetic transition was observed at the range 40-80 GPa, though most of the iron remains in a high-spin state. Thus, the feature observed in Fe₃O₄ is drastically different from the behavior of all other previously studied ferric compounds, therefore requiring further thorough studies and analysis. However, since magnetite contains both Fe²⁺ and Fe³⁺, with Fe³⁺ occupying two different sites, the resulting MS spectra are rather complicated, and a high-quality analysis of the MS data is quite difficult. Comparative studies of different ferric spinels, with Fe²⁺ substituted by another M^{2+} cation, can simplify the analysis of the HP behavior of the postspinel phase. The main objectives of this paper are detailed and accurate combined single-crystal and powder XRD, ⁵⁷Fe MS and electrical resistance studies of the typical normal and inverse $M^{2+}Fe^{3+}{}_{2}O_{4}(M = Mg, Zn, Fe)$ ferric spinels in a wide pressure range, focusing on P > 40GPa, in order to resolve the dilemma of the postspinel structure of ferric spinels (particularly of magnetite), and to clarify the nature of their unusual electronic behavior under pressure.

II. EXPERIMENTAL

Samples of MFe_2O_4 (M = Mg,Zn) were prepared by a solid-state reaction of MgO and ZnO with Fe_2O_3 at high temperatures, as already described [17]. The single-crystal experiments were performed with a sample of $MgFe_2O_4$ and a natural single-crystal sample of magnetite.

Custom 4-pin diamond anvil cells (DACs) made at Bayerisches Geoinstitut and Tel-Aviv University [18], and European Synchrotron Radiation Facility (ESRF) membrane cells [19] along with diamond anvils with culet diameters of $150-320~\mu m$ were used to induce high pressure. Re or stainless steel gaskets were pressed to a thickness of $15-30~\mu m$, and the sample was placed in a cavity of $70-120~\mu m$ diameter. It should be noted that different pressure transmitting media were used for the different measurement techniques. X-ray diffraction measurements were performed using pressurized Ne or He gas, whereas Mössbauer spectra were collected mainly using cryogenically loaded N_2 . For the resistance measurements,

the mixture of Al_2O_3 – NaCl was used both as an insulating layer and a pressure medium. Along with the samples, small spheres of ruby were added for pressure measurement using the ruby R_1 -line fluorescence spectroscopy [20–22]. For $ZnFe_2O_4$, XRD data of a small Au strip was used for manometry [23].

Mössbauer spectroscopy measurements were performed using a nominal 10 mCi 57Co(Rh) point source. Either Wissoft software (WissEl GmbH) or Maestro and MCS-Plus (EG&G Ortec) were used to set the 14.4 keV gate and collect the transmission data from the proportional counter. Low-temperature MS was performed using a custom made top-loading liquid nitrogen/liquid helium cryostat. The typical collection time for each spectrum was ~24 h. A limited number of spectra for ZnFe₂O₄ and Fe₃O₄ were collected using synchrotron energy-domain Mössbauer measurement (SMS) carried out at beamline ID18 at ESRF at temperatures down to 2.4 K (see [24] for more details). These spectra were collected with the source at room temperature and, therefore, are affected by the second-order Doppler shift. The spectra for Fe₃O₄ were collected at ID18 for the same sample which was used for powder XRD studies using He pressure medium. Fitting of the spectra was performed by means of the least-squares method [25] from which the following MS parameters were obtained: relative abundance of the Fe spectral components, the isomer shift (IS), the quadrupole splitting, the hyperfine field, the line-width and an area asymmetry parameter. The reported velocity is with respect to α -Fe at room temperature. The ZnFe₂O₄ and MgFe₂O₄ samples used for MS studies were enriched with 25% ⁵⁷Fe.

Powder XRD measurements were carried out at room temperature in angle-dispersive mode with wavelengths of 0.3738 and 0.4152 Å at ESRF, Grenoble, at the ID27 and ID09 beam lines, respectively. Diffraction images were collected using a MAR CCD detector. The image data were integrated using FIT2D [26,27] and DIOPTAS [28], and the resulting diffraction patterns were analyzed with the GSAS+EXPGUI [29,30] program.

Single-crystal synchrotron XRD experiments have been performed at the Extreme Conditions Beamline (ECB), P02.2, at PETRA III, Hamburg, Germany [31]. Diffraction images were collected using a Perkin Elmer flat panel detector. In the experiment with Fe₃O₄, the energy of the synchrotron radiation was set to $42.72\,\mathrm{keV}(\lambda=0.29023\,\mathrm{\mathring{A}})$, and the x-ray beam was focused down to $3.0(V)\times8.0(H)\,\mu\mathrm{m}^2$ with compound reflective lenses. The single crystal of Fe₃O₄ was compressed from 3 to 51 GPa with a 2 to 5 GPa step. In the experiment with MgFe₂O₄, the synchrotron x-ray beam (42.81 keV, $\lambda=0.28962\,\mathrm{\mathring{A}}$) was focused down to $2.4(V)\times4.0(H)\,\mu\mathrm{m}^2$ using Kirkpatrick-Baez mirrors. The XRD data were collected at four pressure points (2.6, 14.6, 28.4, and 34.4 GPa).

In both experiments, at every pressure step, we collected XRD images during continuous rotation of DACs typically from -25 to +25 on omega. FIT2D [26,27] software was used to analyze the 2D images. CeO₂ standard (SRM 674b) from the National Institute of Standards and Technology was used to calibrate coordinates of the beam center, tilt angle, and tilt plane rotation angle. At selected pressure points, data collection experiments were performed by narrow $0.5\text{-}1^\circ$ omega-scanning in the range from -40° to $+40^\circ$. Integration of the reflection intensities and absorption corrections

were performed using CrysAlisPro software [32]. A single crystal of an orthoenstatite $[(Mg_{1.93}, Fe_{0.06})(Si_{1.93}, Al_{0.06})O_6, Pbca, a = 8.8117(2), b = 5.18320(10), c = 18.2391(3) Å]$ was used to calibrate the instrument model of CrysAlisPro software (sample-to-detector distance, the detector's origin, offsets of the goniometer angles, and rotation of the x-ray beam and the detector around the instrument axis). The structures were solved by the direct method and refined by full matrix least-squares using SHELXTL software [33].

III. RESULTS

A. MS

1. MgFe₂O₄ and ZnFe₂O₄

Mössbauer spectroscopy measurements of MgFe₂O₄ and ZnFe₂O₄ were performed up to \sim 110 and \sim 88 GPa, respectively. At ambient conditions, the Mössbauer spectrum of MgFe₂O₄ consists of two sextets corresponding to two lowpressure (LP) Fe³⁺ structural sites (inverse spinel), tetrahedral and octahedral [Fig. 1(a)]. Both spectral components have almost equal spectral intensity with similar IS values of 0.41(2) mm/s. The two components exhibit different magnetic hyperfine fields $H_{hf} = 50.2(0.32)$ and 52.5(0.32) T at 90 K, typical for a ferric-oxide high-spin ${}^6A_{1g}$ state in tetrahedral and octahedral environments. Between ambient pressure and 20 GPa, the observed spectra barely change with the only observable variation being a slight decrease in IS [see Figs. 1(a) and 2(a)] and about 3% increase in H_{hf} . In the case of ZnFe₂O₄, the Mössbauer spectrum at ambient conditions [Fig. 1(b)] consists of only one doublet, corresponding to Fe³⁺ octahedral sites (normal spinel) with IS = 0.342(6)mm/s and quadrupole splitting QS = 0.425(7)mm/s. As a result of a weak magnetic coupling, ZnFe₂O₄ has a relatively low ordering temperature $T_{\rm N} = 10.5$ K at ambient pressure [34]. Similar to MgFe₂O₄, the spectra barely change with pressure up to 24 GPa [Figs. 1(b) and 2(b)].

At the pressure range 25–40 GPa for both studied materials, a significant change occurs in the MS spectra; namely, the appearance of two new paramagnetic components, and above 40 GPa, the only components remaining are these paramagnetic components. These two new components, designated as PS1 and PS2, are characterized by quite different QS and IS values (Fig. 2). It is noteworthy, that the IS values for the PS1 and LP components are pretty close, suggesting the same nearest-neighbor environment; namely, an octahedral configuration. Low-temperature measurements (Fig. 3) show

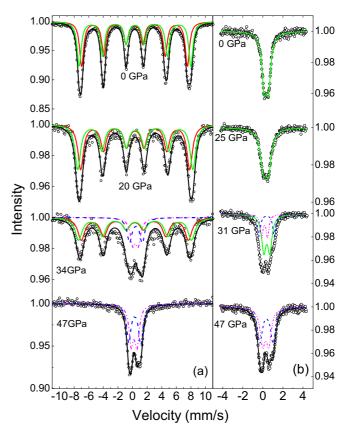


FIG. 1. Mössbauer spectra of (a) MgFe₂O₄ and (b) ZnFe₂O₄ recorded at room temperature and various pressures. For MgFe₂O₄, two Fe³⁺ components at the spinel phase correspond to the tetrahedral and octahedral sites. At 29 GPa, two paramagnetic components appear in the central part of the spectrum. The abundance of the new phase increases with pressure, reaching 100% above 40 GPa. For ZnFe₂O₄, the spectra up to 24 GPa consist of only one doublet, corresponding to Fe³⁺ in the spinel octahedral sites. Above 25 GPa, two new Fe³⁺ sites appear (dashed and dash-dot-dotted lines) corresponding to the components of the HP phase.

that these new components are magnetically split with a Néel temperature of $T_{\rm N}\approx 90$ K (47 GPa) for both materials. The magnetic hyperfine fields $H_{\rm hf}=42.4(3)$ and 46.9(4) T for MgFe₂O₄ at 50 K, typical of a Fe³⁺ HS state with a more covalent nature. Similar $H_{\rm hf}$ values, 40.3(5) and 46.4(3) T at 80 K, were obtained for ZnFe₂O₄.

It is noteworthy that, with the formation of the postspinel phase, the chain of transformations is not finished yet. Low-temperature measurements show that, with further pressure increase above 50 GPa, an appearance of a new broad component (Fig. 4), designated as PS3, is observed in the central part of the spectra, characterized by the significantly reduced IS value of ~0.12 and ~0.17 mm/s for MgFe₂O₄ and ZnFe₂O₄, respectively (Fig. 2). The abundance of the new component increases sluggishly, replacing the PS1 component [see inset Fig. 5(a)]. The temperature dependences of the Mössbauer spectra at 80 GPa for MgFe₂O₄ and 88 GPa for ZnFe₂O₄ are shown in Fig. 4. Low-temperature measurements show the gradual broadening of the absorption spectrum and the increase of QS with the decrease in temperature. Below 50 K, the onset of a magnetic splitting corresponding with the ordering of

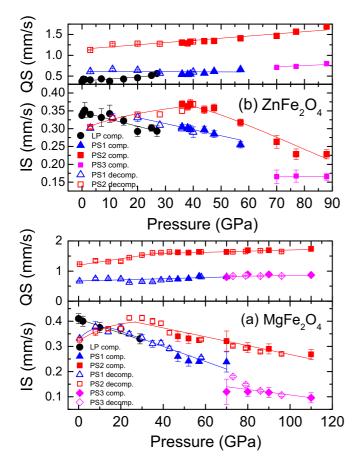


FIG. 2. The pressure evolution of the IS and the quadrupole splitting of (a) MgFe₂O₄ and (b) ZnFe₂O₄ at 300 K relative to a ⁵⁷Co(Rh) source at 300 K. Note that, above 60 GPa, the PS1 component is replaced by the PS3 component characterized by a significantly reduced IS value. All lines are to guide the eye. Symbols: "●" represents the spinel, "▲" and "■" the PS1 and PS2 components, respectively, "◆" the PS3 component. Filled and empty shapes mark compression and decompression cycles, respectively. The pressure uncertainties are 5%.

the HS PS2 component appears, and below 8 K, one clearly observes the onset of a smeared magnetic splitting corresponding with the PS3 component, which are obvious features of a paramagnetic spin relaxation phenomenon (see [35]).

Mössbauer spectra obtained at room temperature during decompression showed a significant hysteresis at the pressure range below 40 GPa, suggesting that the transition to the postspinel phase is not reversible or strongly suppressed by the effect of kinetics. With reducing pressure, only the components of the postspinel phase are observed down to ambient pressure (Fig. 2, and Fig. S1 in the Supplemental Material [36]). Contrarily, no hysteresis was observed in the IS(P) behavior at the range 50–70 GPa, suggesting a fully reversible transformation at this pressure range.

2. Fe₃O₄

In the case of magnetite Fe_3O_4 , the performed low-temperature MS studies show a similar evolution of the MS spectra at the pressure range 45–80 GPa. It is noteworthy that the Fe_3O_4 postspinel phase, at its appearance, is

characterized by three equally abundant Fe HS components $[Fe^{2+}, Fe^{3+}(I), Fe^{3+}(II)]$ [11]. At pressures above 40 GPa, similar to MgFe₂O₄ and ZnFe₂O₄, a new quadrupole-split component, characterized by the significantly reduced IS value of ~ 0.16 mm/s, emerges at the center of the spectra (Fig. 6). This component was designated as $Fe^{3+}(I)_{nm}$. Its abundance increases gradually with pressure at the expense of the Fe³⁺(I) HS component, reaching 1/3 value at ~ 80 GPa when the latter component completely disappears (Fig. 6). However, in contrast to MgFe₂O₄ and ZnFe₂O₄, in Fe₃O₄, the electronic transition is not finished yet with the completion of this transformation. Above 70 GPa, another nonsplit component appears at the center of the spectra, with its abundance increasing gradually at the expense of the Fe²⁺ component (Fig. 6). This component was designated as Fe^{2+}_{nm} . The onset of these new components coincides with the decrease of the magnetic ordering temperature of the HS magnetically split components from 380 K at 55 GPa to 160 K at 82 GPa [11]. It is noteworthy that, in the case of using N₂ pressure medium, part of the Fe²⁺ ions remain in the HS state even at 120 GPa (Fig. 6). Meanwhile, using He as pressure medium results in a complete transformation on Fe²⁺ sites even at about 84 GPa (Fig. 7). We suggest that these observations highlight the sensitivity of the electronic transition to additional stresses and deviation from quasihydrostatic conditions. Low-temperature measurements show no sign of broadening or splitting of the absorption spectrum and no increase of QS for Fe²⁺_{nm} and $Fe^{3+}(I)_{nm}$ components with the decrease in temperature down to 2.4 K, which means no features of a magnetic interaction.

B. XRD

Powder XRD measurements of MgFe₂O₄, ZnFe₂O₄, and Fe₃O₄ were carried out up to 53, 70, and 120 GPa, respectively. Consistent with MS, our XRD studies show an irreversible first-order structural transition at the range 25-40 GPa in each of the spinels. Particularly, the coexistence range of the LP and HP phases for MgFe₂O₄, ZnFe₂O₄, and Fe₃O₄ is 30–42 GPa, 25–36 GPa, and 28–40, respectively. Based on the MS studies, our initial interpretation was that the HP postspinel phases of MFe₂O₄ ferrite spinels are not of the CaMn₂O₄ (Pbcm) or CaTi₂O₄ (Bbmm) structure types. The only structure from the earlier proposed models, characterized by two different Fe³⁺ sites, is the CaFe₂O₄ (*Pnma*) type. Fitting using this structural model gives rather good results in the case of F(calc) weighted (model biased Le Bail) refinement, but poor results in the case of a full-profile Rietveld refinement, which takes into account the atomic positions. Another possibility which should be considered is that, within these structure types, some of the Fe³⁺ cations exchange positions with M^{2+} cations, similar to the inversion within the original spinel structure. In such a case, CaMn₂O₄- or CaTi₂O₄-type structures could be also consistent with the obtained MS results. Indeed, Rietveld refinement of MgFe₂O₄, ZnFe₂O₄, and Fe₃O₄ [37] powder diffraction data, assuming the CaTi₂O₄-type structure with the Fe³⁺ occupying two different crystallographic sites, results in a reasonable fit to the data, as evidenced in Figs. 8(a)-8(c)(background subtracted quality factors: $w_{Rp} < 0.45\%, Rp <$ $0.32\%, w_{Rp} < 1.4\%, Rp < 0.96\%$, and $w_{Rp} < 2.3\%, Rp < 0.96\%$ 1.3%, respectively). However, a fit to CaMn₂O₄-type structure

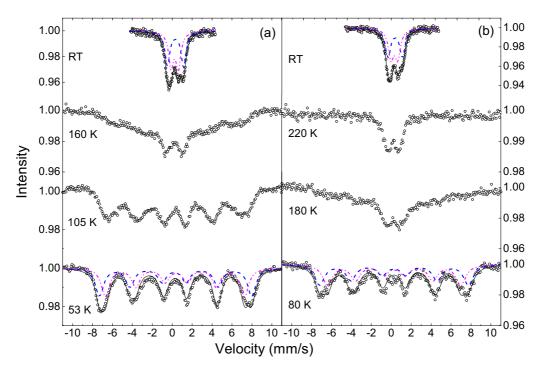


FIG. 3. Mössbauer spectra of (a) $MgFe_2O_4$ and (b) $ZnFe_2O_4$ recorded at 47 GPa and various temperatures; spectra were fit with two Fe^{3+} sites PS2 and PS1 (dashed and dash-dot-dotted lines, respectively). A strong magnetic relaxation effect is observed below 200 K, and below 100 K, the hyperfine fields become saturated. Note a slight discrepancy between the experimental and calculated data at the central part of the spectra at 53 and 80 K.

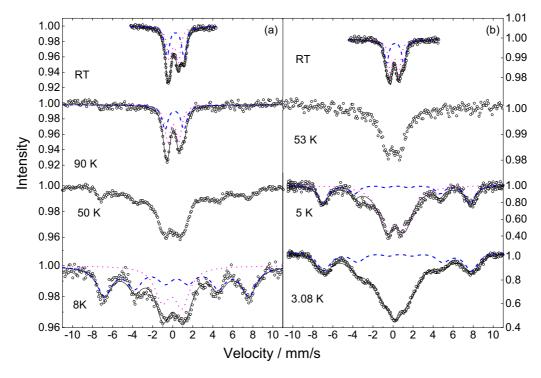


FIG. 4. Mössbauer spectra of (a) MgFe₂O₄ and (b) ZnFe₂O₄ recorded at 80 and 88 GPa, respectively, and various temperatures. The spectra below 50 K are characterized by the two components: the first is characterized by a static magnetic hyperfine interaction $H_{hf} = 45$ T (HS PS2 component, dashed line), and the second by a spin-spin magnetic relaxation spectrum (PS3 component, dotted line), the typical feature of the LS state [31]. Note that the magnetic ordering temperature for the HS component drops from \sim 100 K at 47 GPa (Fig. 3) to below 50 K here. The spectra of ZnFe₂O₄ at 84 GPa, 5 and 3.08 K were collected using energy-domain MS carried out at the beamline ID18 at ESRF.

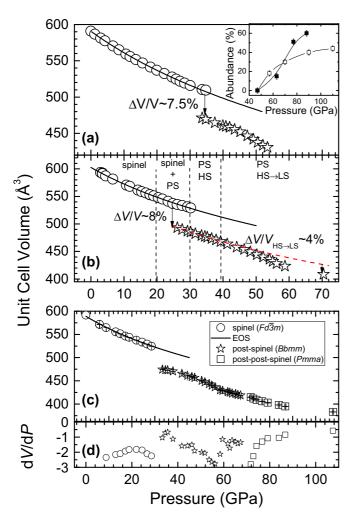


FIG. 5. Unit-cell volume of the spinel and postspinel CaTi₂O₄type phases of (a) MgFe₂O₄, (b) ZnFe₂O₄, and (c) Fe₃O₄ as a function of pressure at room temperature; (d) pressure derivative of the unitcell volume for Fe₃O₄ as a function of pressure. Circles o and stars ☆ represent the spinel and postspinel phases, respectively. The solid lines are fits to the second-order Birch-Murnaghan EOS of the spinel phases extrapolated to \sim 50 GPa, fitting results in the values of $K_0 =$ 170.5(8), 174.1(2) [17], 198(9) GPa, and $V_0 = 591.0(2), 602.8(4)$, and 589.0(2) Å³ for M = Mg, Zn, and Fe, respectively. The thirdorder EOS for Fe₃O₄ results in $V_0 = 590(3) \text{ Å}^3$, $K_0 = 182(29) \text{ GPa}$, and K' = 5(2). The dashed line represents a fit to the postspinel phase of ZnFe₂O₄ at the pressure range 25-40 GPa. Note a steeper decrease of unit-cell volume with pressure increase above 40 GPa, indicating a HS-LS transition on the octahedral sites. The inset shows the abundance of the quadrupole-split paramagnetic MS component as a function of pressure in MgFe₂O₄ and ZnFe₂O₄ (open and solid squares, respectively).

with the Fe³⁺ occupying two different crystallographic sites is also possible, with rather similar results.

To resolve this problem and to get a final structure assignment, we have performed single-crystal XRD studies of MgFe₂O₄ and Fe₃O₄. For both materials, the HP phase has been identified as belonging to the *Bbmm* (or equivalent *Cmcm*) space group (Fig. 9). The complete details of the single-crystal refinement for MgFe₂O₄ at 27.4(5) GPa and

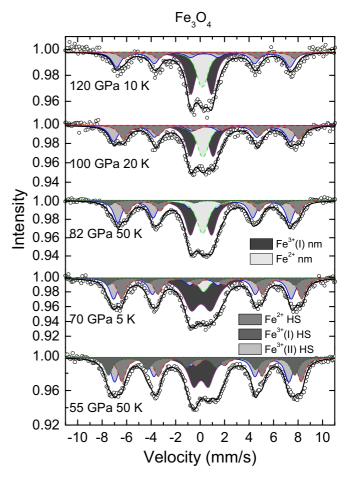


FIG. 6. Mössbauer spectra of Fe_3O_4 recorded at various pressures and low temperatures (N_2 pressure medium). Note that, at its appearance, the postspinel phase is characterized by three equally abundant Fe components [14] $[Fe^{2+}, Fe^{3+}(I), Fe^{3+}(II)$: red dashed, olive dotted, and blue solid lines, respectively]. Above 40 GPa, an onset of a new quadrupole-split $Fe^{3+}(I)_{nm}$ component is observed (magenta dotted line), replacing the $Fe^{3+}(I)$ component. Around 70 GPa, another unsplit Fe^{2+}_{nm} component appears (green dashed line), gradually replacing the Fe^{2+} component. Components are shaded for grayscale viewing offline.

Fe₃O₄ at 28.4(5) GPa are given in Table I and Supplemental Table 1 [36], respectively. The obtained unit-cell volumes are similar to those obtained in previous experiments [10,38] reporting either the CaTi₂O₄ or CaMn₂O₄ structures and similar to our powder XRD data. However, since the singlecrystal refinement is free from additional effects such as texture, compared to powder diffraction refinement, we are able to more accurately determine the atomic positions and site occupancy of the different cations. Table II shows the atomic positions and site occupancies of the Fe^{3+} , Mg^{2+} , and O² ions of MgFe₂O₄. Thus, from single-crystal refinement, we determine that the Fe³⁺ cations do not occupy solely the octahedral 8f Wyckoff sites; there is in fact a partial inversion with the Mg²⁺ cations: about 64% of Fe³⁺ ions occupy octahedral sites, while the rest are in bicapped trigonal prism. For the Fe₃O₄ sample (see Supplemental Table 2 [36]), we cannot determine whether there is any inversion because

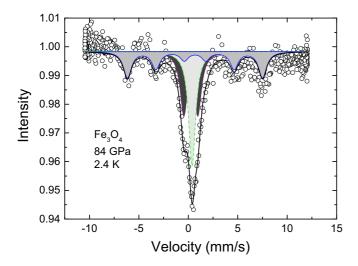


FIG. 7. The spectra of Fe_3O_4 at 84 GPa and 2.4 K collected using energy-domain MS (He pressure medium). Note that only three equal abundant components are observed at this pressure, namely Fe^{2+}_{nm} , $Fe^{3+}(I)_{nm}$, and $Fe^{3+}(II)$. This suggests that using the He pressure medium results in a complete electronic transformation on the octahedral sites at about 84 GPa. Meanwhile, Fe^{3+} ions occupying square antiprism coordinated sites remain in the HS state (blue solid line). The lack of any sign of a magnetic interaction for Fe^{2+}_{nm} and $Fe^{3+}(I)_{nm}$ components (green dashed and magenta dotted lines, respectively) above 84 GPa and down to 2.4 K suggests that these components emanate not from a paramagnetic, but rather from a nonmagnetic state. Components are shaded for grayscale viewing offline.

it is not possible to distinguish between Fe²⁺ and Fe³⁺ with single-crystal diffraction. As expected, the transition from a structure with four- and sixfold coordination to a structure with six- and eightfold coordination results in an increase in the average $\langle \text{Fe-O} \rangle$ distances. Within the spinel phase of MgFe₂O₄, at 28.4 GPa, the average $\langle \text{Fe-O} \rangle$ distances are 1.814(4) and 1.967(2) Å for the tetrahedral and octahedral sites, respectively, whereas within the *Bbmm* phase, at 28.4 GPa, the $\langle \text{Fe-O} \rangle$ distances are 1.963(2) and 2.176(2) Å for the octahedral and bicapped trigonal prism sites, respectively. Even taking into account only the six shortest Fe-O bonds within the higher coordinated site $\langle \text{Fe-O} \rangle = 2.038(3)$, which is still significantly larger than observed within the spinel phase.

Thus, we can conclude that all studied ferrite spinels undergo, at the range 25–40 GPa, a first-order structural phase transition to the postspinel $CaTi_2O_4$ -type structure. This structural transition is accompanied by about 8% volume reduction. Another noticeable change in the V(P) data, namely a steeper decrease of unit-cell volume with pressure increase, occurs at higher pressures, above 42, 40, and 38 GPa for MgFe₂O₄, ZnFe₂O₄, and Fe₃O₄, respectively (Fig. 5). This volume decrease is seen in ZnFe₂O₄ and Fe₃O₄ up to \sim 60GPa, indicating a continuous increase in compressibility and furthermore showing signs of V(P) stabilization between 60 and 70 GPa. Such V(P) behavior suggests a significant electronic transformation taking place within the HP phase. Correspondingly, the V(P) data for the postspinel phase at this pressure range could not be fit with a "standard" equation

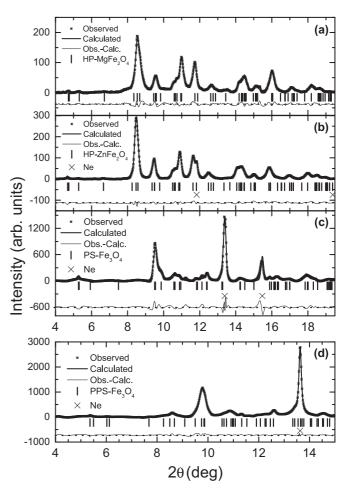


FIG. 8. Example of Rietveld refinement of the postspinel phase of (a) MgFe₂O₄ at 50.5(2) GPa, (b) ZnFe₂O₄ at 51.0(5) GPa, and (c) Fe₃O₄ at 63.6(8) GPa and RT assuming a CaTi₂O₄-type structure (SG *Bbmm*), and (d) F(calc) weighted (model biased) fitting of Fe₃O₄ at 79(1) GPa, assuming a *Pmma* post-postspinel type structure. Asterisks * mark the measured data, the solid line is the best fit to the data, and the dashed line shows the difference between the observed and calculated intensities (all background subtracted). Horizontal lines and X's show the expected diffraction peaks of the sample and neon, respectively. Here, $\lambda_{a,b} = 0.3738$ and $\lambda_{c,d} = 0.4152$ Å.

of state (EOS). Meanwhile, to define an EOS for the HP phase before the steeper decrease of V, one can use the data at the pressure below 40 GPa. Thus, for $ZnFe_2O_4$, the

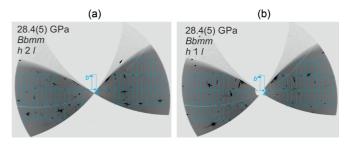


FIG. 9. Reciprocal space reconstructions of HP-Fe₃O₄ at 28.4(5) GPa corresponding to (a) h 2 l and (b) h 1 l planes, respectively. Sites on a blue lattice represent predicted positions of the reflections. Blue circles show systematic absences corresponding to B centering.

TABLE I. Details of crystal structure refinements of Bbmm-Fe₂MgO₄ (27.4 GPa).

Crystallographic data	$\mathrm{Fe_2MgO_4}$	
Crystal system	Orthorhombic	
Space group	Bbmm	
a (Å)	9.251(6)	
b (Å)	9.380(4)	
c (Å)	2.7583(6)	
α (°)	90	
β (°)	90	
γ (°)	90	
$V(\text{Å}^3)$	239.37(19)	
Z	4	
F(000)	384	
θ range for data collection (°)	2.520 to 10.427	
Completeness to $d = 0.8 \text{ Å}, \%$	0.633	
Index ranges	-8 < h < 9	
	-10 < k < 11	
	-3 < l < 3	
Reflections collected	210	
Independent reflections/ $R_{\rm int}$	101/0.0425	
Refinement method	Full matrix least squares on F^2	
Data/restraints/parameters	101/1/19	
Goodness of fit on F^2	1.073	
Final R indices $[I > 2\sigma(I)]$, R_1/wR_2	0.0412/0.0990	
Type = "Periodical" R indices (all data), R_1/wR_2	0.0514/0.1104	
Largest difference peak/hole $(e/\text{Å}^3)$	0.624/-0.815	

data for the molar volume were fit using the second-order Birch-Murnaghan (BM2) EOS [39]. The performed fit results in $K_0 = 134(18)$ GPa and $V_0 = 571(10)$ Å³, where K_0 and V_0 are the bulk modulus and the unit cell volume at 1 bar and 300 K, respectively. Extrapolation of the obtained EOS to higher pressures allows us to estimate a volume reduction corresponding with the proposed electronic transition. This estimation gives $\Delta V/V_0 \approx 4\%$.

It is noteworthy that, for Fe₃O₄ at pressures above 65 GPa, we observed, similar to [38], the appearance of new diffraction peaks in the XRD patterns, whose intensities start to be significant above 70 GPa (Fig. 10). The changes observed in the patterns are caused by a second structural transition to a post-postspinel phase, which is completed above 72 GPa. However, the obtained patterns differ from those observed

by Ricolleau *et al.* [38], which could be related with different experimental conditions (in [38], XRD patterns were collected after annealing at temperatures up to 2300 K). In our case, the obtained patterns could be fitted well with the *Pmma* structure. A model biased F(calc) weighted refinement of the diffraction data results in a reasonable fit to the data, as evidenced in Fig. 8(d) ($w_{Rp} < 0.5\%, Rp < 0.4\%$).

Pmma is a nonisomorphic subgroup of *Bbmm*, which loses the *C*-centered symmetry, and the structural transition from *Bbmm* to *Pmma* belongs particularly to the order-disorder transition [40]. Correspondingly, no volume change could be expected in such a case. In agreement with this, our V(P) data do not show any appreciable volume change at the transition [Fig. 5(c)]. It is noteworthy that, for Fe₃O₄, one can observe an additional kink on the V(P) curve at the

TABLE II. Atomic coordinates and equivalent isotropic displacement parameters for MgFe₂O₄.^a

Atom	Wyckoff site	x	У	Z	$U_{\rm eq}^{\rm b}$ or $U_{\rm iso}^{\rm c}({\rm \AA}^2)$	Site occupancy
Fe1	8 <i>f</i>	0.1306(2)	0.07245(18)	0.0000	0.0191(8)	0.6360(15)
Mg1	8f	0.1306(2)	0.07245(18)	0.0000	0.0191(8)	0.3640(15)
Fe2	$\overset{\circ}{4c}$	0.3815(3)	0.2500	0.0000	0.0209(9)	0.728(3)
Mg2	4c	0.3815(3)	0.2500	0.0000	0.0209(9)	0.272(3)
01	4b	0.5000	0.0000	0.0000	0.021(2)	1
O2	4c	0.0306(13)	0.2500	0.0000	0.019(2)	1
O3	8 <i>f</i>	0.2244(9)	-0.1206(7)	0.0000	0.0179(14)	1

 $^{^{\}mathrm{a}}U_{\mathrm{eq}}$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

^bFor metal atoms refined in anisotropic approximation.

^cFor oxygen atoms refined in isotropic approximation.

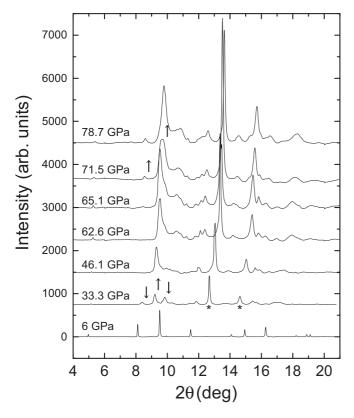


FIG. 10. Powder XRD patterns of Fe₃O₄ at various pressures during compression. At 33 GPa, * mark the Ne pressure medium peaks, \downarrow mark the disappearance of the spinel phase, and \uparrow mark the appearance of the *Bbmm* postspinel phase. At 71.5 GPa, \uparrow mark the appearance of the post-postspinel *Pmma* phase. Here, $\lambda = 0.4152 \text{ Å}$.

range 70–85 GPa characterized by a steeper volume decrease [Fig. 5(d)].

C. Resistance measurements

Here, R(P,T) measurements for MgFe₂O₄ were carried out during compression to 81 GPa followed by decompression to ambient pressure, and for ZnFe₂O₄ during compression to 92 GPa. The pressure dependence of the resistance measured at RT is shown in Fig. 11. As can be seen in the spinel phase, R(P) shows a minor increase with pressure increase reaching a maximum at \sim 25 GPa, where the postspinel phase appears. Above that, it drops continuously, demonstrating a distinctive feature at the 45–70 GPa range, namely a more rapid decrease. For ZnFe₂O₄, a more rapid decease of resistance was observed also in the 25-35 GPa spinel-postspinel phase coexistence range. It is noteworthy that, up to the highest pressures measured, MgFe₂O₄ and ZnFe₂O₄ remain semiconductors, showing a negative R(T) slope [Fig. 11(a), inset]. At a fixed pressure, the resistivity data fairly accurately obey the relation of $lnR = lnR_0 + E/k_BT$, where k_B is the Boltzmann constant, E the electrical transport activation energy [Fig. 11(b), inset]. At the 45-85 GPa range, the value of E decreases from 0.17 to 0.11 and 0.07 eV for MgFe₂O₄ and ZnFe₂O₄, respectively. An approximation of the E(P) data to highest pressures suggests a possible closing of the gap only at about 160 and 115 GPa, correspondingly.

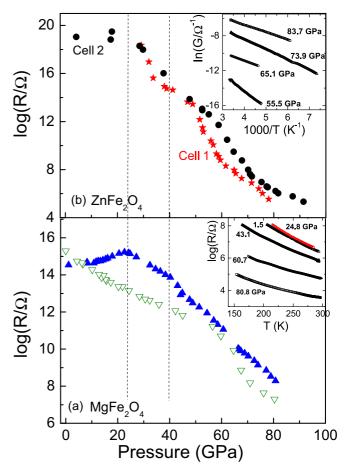


FIG. 11. The pressure dependence of the resistance of (a) $MgFe_2O_4$ and (b) $ZnFe_2O_4$ recorded at 295 K. The solid (\blacktriangle) and open (\bigtriangledown) triangles are data points obtained during the cycles of compression and decompression, respectively, of the $MgFe_2O_4$ sample. Note the minor increase of R up to 25 GPa in the spinel phase and more rapid decrease at the range 45–70 GPa. The vertical dashed lines set the boundary of the coexistence between the spinel and postspinel phases. The inset shows R(T) and conductance G(T) of the samples at various pressures.

IV. DISCUSSION

Summarizing the obtained MS and XRD data, we can conclude that, for all the studied materials, a pressure-induced first-order structural phase transition takes place at the pressure range 25-40 GPa. This phase transition is manifested by the observed changes in XRD patterns and dramatic changes of the hyperfine interaction parameters. As in Fe₃O₄, the postspinel phase of MFe₂O₄ ferrites comprises two distinct Fe³⁺ sites characterized by different QS and IS values. Analysis of the powder XRD data allows assuming for the postspinel phase a CaTi₂O₄- or CaMn₂O₄-type structure. Single-crystal XRD data point at the CaTi₂O₄-type (Bbmm) as the postspinel structure in $M\text{Fe}_2\text{O}_4$ (M = Mg, Zn, Fe) ferric spinels. Taking into account that two new MS spectral components, PS1 and PS2, are characterized by rather different QS and IS values, we can propose that the ferric ions occupy in this structure different crystallographic sites. This corresponds well with two cationic sites in the CaTi₂O₄-type structure, namely an octahedral and a bicapped trigonal prism, characterized by six- and eightfold coordination polyhedra, respectively. Such different Fe $^{3+}$ neighborhoods should result in rather different hyperfine parameters as was actually observed. It is noteworthy that the IS values for the PS1 and LP components at about 35 GPa are pretty close, while the IS value for the PS2 component is 0.065(14) and 0.084(28) mm/s higher for ZnFe $_2$ O $_4$ and MgFe $_2$ O $_4$, respectively, suggesting an octahedral environment, similar to the LP phase, for the PS1 component and a higher coordination number, namely bicapped trigonal prism for PS2.

In the studied ferrites, immediately following the phase transition, Fe³⁺ in both sites is in the high-spin state. However, with further pressure increase, a steeper decrease of unit-cell volume and lattice parameters is seen at the pressure range 40-70 GPa for all studied ferrites. This unusual behavior of the unit-cell volume corroborates with an appreciable change in the features of the MS spectra of MgFe₂O₄ and ZnFe₂O₄; namely, a new quadrupole-split paramagnetic component appears above \sim 50 GPa, with abundance increasing sluggishly with pressure [Fig. 5(a), inset]. This new PS3 component is characterized by a significantly reduced IS value, which corresponds to an increase of the effective s density at the nucleus. The new quadrupole-split component can result from either (i) a HS-LS transition, or (ii) a correlation breakdown (a Mott transition) in which a magnetic interaction vanishes [15]. In the present case, at low temperatures, the PS3 component exhibits a smeared magnetic splitting, which is a feature of a weak magnetic interaction, a paramagnetic spin relaxation phenomenon [34]. Thus, the appearance of the features of a paramagnetic spin relaxation phenomenon at low temperatures and the absence of a corroborating insulator-to-metal transition at this pressure range allows us to conclude that the HS-LS transition takes place in the present case. Since MS observations refer to the PS1 component, as undergoing a spin crossover, we conclude that the transition to the LS state originates only on the octahedral sites. Such a transition results in shorter Fe-O distances, hence in a reduced crystal volume and lattice parameters, which were clearly observed in XRD experiments (Fig. 5). It is noteworthy that the ~4% volume decrease, corresponding with the HS-LS transition in ZnFe₂O₄ ZnFe₂O₄ [see Fig. 5(b)], suggests about 12% decrease of the Fe³⁺ octahedral volume, which agrees well with results of previous studies [41,42]. In the case of MgFe₂O₄, according to single-crystal XRD data, about 64% of Fe³⁺ occupies octahedral sites; in the meantime, the abundance of the LS MS component reaches only ~44% at 110 GPa [see Fig. 5(a), inset]. It means that, in MgFe₂O₄, the HS-LS transition on the octahedral sites is not finished even at such

According to the XRD and MS data, the HS-LS transition is sluggish: an anomalous crystal volume behavior is observed at the range 40–70 GPa and is attributed to the interplay between normal compressibility and spin variation effect on the Fe³⁺ ionic radius. Another interesting finding is a significant decrease of the *Néel* temperature $T_N(P)$, despite an expected increase in the exchange and super-exchange interactions with density increase. This is also a consequence of the HS-LS transition on the octahedral sites, which excludes in fact nearly half of the Fe³⁺ ions from the super-exchange

interaction. As a result, in both systems, T_N drops from \sim 90 K at 47 GPa to \sim 50 K at 80 GPa.

Thus, in contrast to other studied ferric oxides (see [15,16]), in the case of postspinel structures of MgFe₂O₄ and ZnFe₂O₄, about half of the Fe³⁺ ions remain in the HS state at least up to 110 GPa, the highest pressure achieved in this paper. Another special feature distinguishing the spinel and postspinel structures is the absence of any first-order pressureinduced electronic transitions typical of Fe³⁺ at these extreme conditions; namely, no Mott transition takes place as verified by the negative slope of R(T) up to \sim 95 GPa and the presence of magnetism. Furthermore, in MgFe₂O₄, no gap closure is expected up to 160 GPa. Also, no first-order HS-LS transition, typical for ferric oxides, was observed, only the sluggish spin crossover on the octahedral sites. It is noteworthy that recent studies of α -LiFeO₂, crystallizing in the rock salt structure with random distribution of Li and Fe ions, shows very similar features of the pressure-induced spin crossover, namely a sluggish second-order Fe³⁺ HS-LS transition, which starts at 50 GPa and is not completed even at \sim 100 GPa [43]. It was proposed that the observed feature of the HS-LS transition in α -LiFeO₂ is caused by the random environment of Fe³⁺ ions [41]. The random environment of Fe³⁺ ions is expected also in the present ferric spinels. Hence, the presented features of the spin crossover in the studied spinels could serve as another confirmation of the effect of the nearest Fe³⁺ environment upon the transition features.

Thus, these postspinels remain, up to the highest pressures measured, semiconducting and paramagnetic with a mixed spin state: S = 5/2 and S = 1/2 for Fe³⁺ populating eightcoordination and six-coordination sites, respectively. Recent studies of Fe₂O₃, hematite [44], provide experimental and theoretical evidence for a possible site-selective multistage correlation breakdown. They show that, in a material with a complex crystal structure, containing transition metal cation(s) in different environments (i.e. different coordination numbers), Mott insulator-to-metal transition and collapse of magnetism does not necessarily occur simultaneously and may propagate through different crystallographic sites at different degrees of compression. In that case, a structural transition to a crystal structure with multiple crystallographic sites, with octahedral and eightfold coordinated sites, results in metallization and collapse of magnetism only in half of the Fe³⁺ (those in octahedral sites), while Fe³⁺ in the higher coordinated sites remain HS up until a second structural phase transition results in a single octahedral site for all Fe³⁺. Our present studies of MgFe₂O₄ and ZnFe₂O₄ postspinels demonstrate similar features also for a HS-LS transition. Such a site-selective spin transition was recently discovered in (Mg,Fe)(Si,Fe)O₃ perovskite, where according to both experiments [45] and DFT + U calculations [46], Fe³⁺ in octahedral sites undergoes the transition to low-spin in the range of 40-70 GPa, while Fe³⁺ in dodecahedral sites remains high-spin.

Similar conclusions could be reached also for Fe_3O_4 , magnetite. However, in the case of Fe_3O_4 , where $Mg^{2+}(Zn^{2+})$ is replaced by Fe^{2+} , the observed electronic transformation is even more complicated. Indeed, the performed low-temperature MS studies of magnetite show an evolution of the MS spectra at the pressure range 40–60 GPa, which is similar to the other studied ferrites. As was mentioned above, Fe_3O_4 postspinel phase,

at its appearance, is characterized by three equally abundant HS Fe components $[Fe^{2+}, Fe^{3+}(I), Fe^{3+}(II)]$ [11]. Taking into account the significantly lower IS value obtained for the Fe³⁺(I) component, compared to the Fe³⁺(II) component (see Fig. 6), one can suggest an octahedral environment for the $Fe^{3+}(I)$ sites and bicapped trigonal prism for the $Fe^{3+}(II)$. At pressures above 40 GPa, similar to MgFe₂O₄ and ZnFe₂O₄, a new quadrupole-split component Fe³⁺(I)_{nm}, characterized by a significantly reduced IS value, emerges at the center of the spectra, gradually replacing the Fe³⁺(I) component (Fig. 6). Correspondingly, $T_{\rm N}$ drops to 160 K at 82 GPa. Coinciding with the onset of this MS component, the steeper decrease of unit-cell volume is observed above 38 GPa followed by the V(P) stabilization above 57 GPa. This suggests that, similar to MgFe₂O₄ and ZnFe₂O₄, a second-order HS-LS transition takes place on the octahedral Fe³⁺(I) sites. In Fe₃O₄, this electronic transition is followed by the structural transition from Bbmm to Pmma structure, which could be considered as the order-disorder transition [40]. In a similar transition in Fe₂TiO₄ [40], the octahedral cationic sites of the Bbmm structure are split into two crystallographic nonequivalent sites in post-postspinel Pmma, where Fe²⁺ and Ti⁴⁺ are ordered on the sixfold coordination sites 4i and 4j, respectively. In the case of Fe₃O₄, the low-spin Fe³⁺ ion, characterized by an extremely small ionic radius compared to HS Fe²⁺, plays the role of Ti⁴⁺, initiating obviously the transition to the *Pmma* structure, where LSFe³⁺ and HSFe²⁺ are ordered on the sixfold coordination sites similar to Fe₂TiO₄.

It is noteworthy that, in contrast to $MgFe_2O_4$ and $ZnFe_2O_4$. where the Fe³⁺ HS-LS transition on the octahedral sites does not result in a closure of the insulating gap, in the case of Fe₃O₄, one can observe a clear trend towards a gap closure above 50 GPa [11]. Upon its inception, at \sim 25 GPa, the HP phase of Fe₃O₄ is a semiconductor with relatively strong carrier localization [11], but corroborating with the secondorder HS-LS transition of ferric ions on the octahedral sites, an incipient semimetal behavior is observed, characterized by rather complicated R(T) features (see Fig. 12). With further pressure increase, an onset of the new nonsplit Fe²⁺_{nm} component in low-temperature MS spectra is observed around 70 GPa (Fig. 6) following the structural transition to *Pmma* phase above 65 GPa and corroborating with a steeper decrease of unit-cell volume at the pressure range 70-85 GPa. This signifies further propagation of the gradual collapse of magnetism corroborating with a continuous sluggish metallization process, which results in a distinct metallic behavior above 90 GPa (Fig. 12). We note that a recent study of the electrical transport properties of Fe₃O₄ by Muramatsu et al. [47], despite qualitative agreement with our data, did not detect an onset of positive R(T) slope up to 98 GPa. In this paper, such a slope occurs at $P \ge 99$ GPa. What is clear from their study is, similar to this paper, the tendency towards gap closure above 60 GPa, with the resistance changing by a factor of \sim 2 from RT down to a few Kelvin above 64 GPa, signifying that the sample is in fact no longer semiconducting. The difference between the results of both of these studies could be related to the sensitivity of the appearance of the metallic phase to stress conditions (e.g. Daphne oil vs Al₂O₃/NaCl pressure medium), sample stoichiometry (see sample synthesis details in [48]), and the details of the measurement technique (e.g. contact

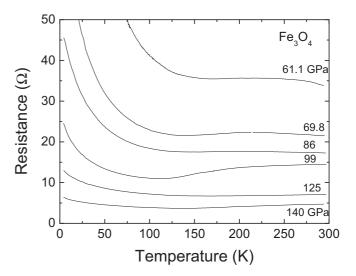


FIG. 12. The temperature dependence of the resistance of Fe_3O_4 at various pressures (part of these data were published previously in [14]). An incipient semimetal behavior is observed above 60 GPa followed by a sluggish metallization above 86 GPa.

resistance, direct contact between sample and diamond-anvil, maximum applied electric current). It is noteworthy that, in He pressure medium, the transformation HS $Fe^{2+} \rightarrow Fe^{2+}_{nm}$ is completed already at about 85 GPa, and furthermore, no appreciable change of the MS spectra is observed, while in the N₂ pressure medium, the transition is not completed even at 120 GPa. The lack of any sign of a magnetic interaction for Fe^{2+}_{nm} and $Fe^{3+}(I)_{nm}$ components above 84 GPa and down to 2.4 K (Fig. 7) prompted us to conclude that these components emanate not from a paramagnetic, but rather from a nonmagnetic state [49]. All these features, namely the loss of the magnetic interactions corroborating with the onset of metallic state, are typical features of the correlation breakdown in strongly correlated systems [15,50,51]. However, in the case of Fe₃O₄, one can see that the correlation breakdown takes place only in the octahedral sublattice, while a third of the iron cations, namely Fe³⁺(II) sites, remain in the HS state. As it was mentioned above, a similar site-selective Mott transition was observed recently in Fe₂O₃, where a partial collapse of magnetism concurrent with an insulator-to-metal transition takes place around 50 GPa, while a complete collapse of magnetism and correlation breakdown was observed only at \sim 70 GPa. In Fe₃O₄, the process is even more complicated. The second-order HS-LS transition on the Fe³⁺ octahedral sites, starting around 40 GPa, corroborates with a clear trend to gap closure and formation of a semimetal state above 50 GPa. Above 70 GPa, a sluggish metallization process also is further observed, corroborating with the onset of the nonmagnetic Fe²⁺ component. This very sluggish metallization process, corroborating with the gradual collapse of magnetism, presumably only on the octahedral sites, is a characteristic feature of the multistage correlation breakdown at the present case. However, even at 120 GPa, the Mott transition in Fe₃O₄ is not completed yet since Fe³⁺(II) still remain in the HS state. It is noteworthy that the comparison with two other studied spinels allows us to propose a crucial role of the substitution of Mg/Zn in the octahedral sites by Fe²⁺ ions, which results in the closure of the gap at relatively LPs. In two other materials (MgFe₂O₄ and ZnFe₂O₄), the gap closure is expected only above 115 GPa, and obviously, because of the complex crystal structure, containing Fe³⁺ in different environments, this will be only the first stage of the multistage process. It is noteworthy that a recent publication of Rahman et al. [52] claims a spin crossover and the onset of the Mott insulator-metal transition (IMT) in the mixed Zn_{0.2}Mg_{0.8}Fe₂O₄ spinel, corroborating with the transition to a postspinel phase around 20 GPa. This claim was based on the disappearance of all the Raman modes above 33 GPa and the absence of the resistivity variation at the temperature range 300-420 K. However, taking into account the results obtained in this paper for the pure MgFe₂O₄ and ZnFe₂O₄ spinels, the claimed IMT and a collapse of Fe³⁺ magnetic moments at such relatively LPs raises many questions and requires, obviously, further verification (for details, see Fig. S2 in the Supplemental Material [36]).

V. CONCLUSIONS

In conclusion, our studies reveal for all studied ferric spinels, $M\text{Fe}_2\text{O}_4$ (M=Mg,Zn,Fe), at the pressure range 25–40 GPa, an irreversible first-order structural transition to the postspinel CaTi_2O_4 -type structure. The postspinel phase is characterized by a partial inversion of the orthorhombic *Bbmm* structure and associated Fe \leftrightarrow M site exchange, where about half of Fe³⁺ enters eightfold coordinated polyhedral sites. In all studied postspinel phases, a second-order HS-LS transition is observed on the octahedral Fe³⁺ sites at around the range of 40–60 GPa, while Fe³⁺ cations occupying bicapped trigonal prism sites remain in the HS state. The observed features of the pressure-induced HS-LS transition in the studied spinels serve as further evidence of the effect of the nearest Fe³⁺ environment upon the spin crossover features [43].

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The HS-LS transition is accompanied by an appreciable resistance decrease, but MgFe₂O₄ and ZnFe₂O₄ remain semiconducting up to \sim 90 GPa, and the gap closure is expected only above \sim 115 GPa. Contrarily, in the case of Fe₃O₄, the sluggish HS-LS transition on the Fe³⁺ octahedral sites corroborates with a clear trend to a gap closure and formation of a semimetal state above 50 GPa. Above 70 GPa, a gradual collapse of magnetism corroborating with a sluggish metallization process further extends capturing also Fe²⁺ sites, while about half of Fe³⁺ sites still remain in the HS state. These electronic transformations in Fe₃O₄ coincide with another structural phase transition to the post-postspinel *Pmma* structure. All these features are characteristics of the multistage correlation breakdown process, confirming the concept of a site-selective Mott transition proposed recently in [44].

ACKNOWLEDGMENTS

A few of the SXRD data were collected at the ID-09 beamline of the ESRF, Grenoble, France. We are grateful to Michael Hanfland at the ESRF for providing assistance in using beamline ID-09. We thank A. Melchior and R. Arielly for assisting with the powder XRD measurements. We thank S. Layek and M. Shulman for assisting with the SMS measurements. We thank A. Motil for his assistance with analysis of related XRD data. We are grateful to Ulf Hålenius (Swedish Museum of Natural History, Stockholm, Sweden) for providing the single crystal of MgFe₂O₄. This research was supported by the Israeli Science Foundation under Grant No. 1189/14. The authors are grateful to A. Kurnosov for his assistance with the Raman spectroscopy measurements.

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