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Transport properties of cubic crystalline $\text{Ge}_2\text{Sb}_2\text{Te}_5$: A potential low-temperature thermoelectric material

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$\text{Ge}_2\text{Sb}_2\text{Te}_5$ (GST) has been widely used as a popular phase change material. In this study, we show that it exhibits high Seebeck coefficients $200\text{--}300\text{ }\mu\text{V/K}$ in its cubic crystalline phase (*c*-GST) at remarkably high *p*-type doping levels of $\sim 1 \times 10^{19}\text{--}6 \times 10^{19}\text{ cm}^{-3}$ at room temperature. More importantly, at low temperature ($T = 200\text{ K}$), the Seebeck coefficient was found to exceed $200\text{ }\mu\text{V/K}$ for a doping range of $1 \times 10^{19}\text{--}3.5 \times 10^{19}\text{ cm}^{-3}$. Given that the lattice thermal conductivity in this phase has already been measured to be extremely low ($\sim 0.7\text{ W/m-K}$ at 300 K) [K. S. Siegrist *et al.*, Rep. Prog. Phys. **78**, 013001 (2015)], our results suggest the possibility of using *c*-GST as a low-temperature thermoelectric material. © 2015 AIP Publishing LLC.

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Low-temperature thermoelectric (TE) materials are important for refrigeration applications in electronics, infrared detectors, computers, and other areas.² However, in contrast to high temperature TE materials, there is relatively little progress in the discovery of high performance low temperature TE materials. The most widely used materials at ambient temperature and below are $(\text{Bi,Sb})_2\text{Te}_3$ derived compounds.^{3,4} These have figures of merit, ZT , of roughly unity at 300 K with lower ZT values at lower temperatures. Here, we show that cubic $\text{Ge}_2\text{Sb}_2\text{Te}_5$ (GST) phase change material has potential to become an excellent thermoelectric material in the important temperature range from 200 K to 300 K . Specifically, we find using Boltzmann transport calculations that disordered cubic *p*-type GST will have high thermopowers in the range of $200\text{ }\mu\text{V/K--}300\text{ }\mu\text{V/K}$ at remarkably high doping levels consistent with good conductivity. GST already is known to have a very low thermal conductivity.¹

Pseudobinary alloys of $(\text{GeTe})_m\text{--}(\text{Sb}_2\text{Te}_3)_n$ have been used in data storage for more than a decade because of their fast phase switching between metastable crystalline (cubic) and amorphous phases.^{5–7} The above room temperature thermoelectric properties of $(\text{GeTe})_m\text{--}(\text{Sb}_2\text{Te}_3)_n$ were previously studied experimentally.^{8–12} In particular, for $m = 2$ and $n = 1$; ordered $\text{Ge}_2\text{Sb}_2\text{Te}_5$ was reported to show interesting thermoelectric properties, even at room temperature.^{9,11,12}

Disorder in the crystal structure can be beneficial for thermoelectric properties^{13,14} and hexagonal GST has been already reported to have reasonably good thermoelectric properties.^{8,15} The disorder in phase change materials was discussed in detail by Zhang and co-workers.¹⁶ We therefore investigate the thermoelectric properties of cubic disordered GST. Previous measurements of thermoelectric properties in *c*-GST were restricted to thin-film samples, where the thermoelectric properties were found to vary significantly

depending on the thickness of the sample.^{11,17} The thermoelectric properties of *c*-GST in the bulk form have remained unknown. We note that Lee and co-workers have discussed resonant bonding in the context of thermoelectric performance¹⁸ and that resonant bonding has been identified in phase change materials.^{19,20}

The *ab-initio* calculations presented here were carried out using the linearized augmented plane-wave (LAPW) method²¹ as implemented in the WIEN2K code.²² The structures were relaxed using the Perdew-Burke-Ernzerhof (PBE) type functional within the generalized gradient approximation (GGA) then a semilocal functional of Tran and Blaha (TB-mBJ)²³ was employed to treat the exchange-correlation potential in the calculations of electronic, optical, and transport properties. This functional form was shown to better describe the band gaps of a variety of semiconductors and insulators.^{24,25}

The cubic phase of GST occurs in a rock-salt structure with Te occupying one site and Ge, Sb, and vacancies randomly occupying the other site. We generated special quasi-random structures (SQSs) using the prescription of Zunger *et al.*²⁶ for our first principles study. The SQSs were generated such that all the pair correlation functions (PCFs) up to the next nearest neighbors are identical to the average PCF of an infinite random structure. We used the mcsqs code as implemented in ATAT software package²⁷ to generate the SQSs using a Monte Carlo algorithm. We considered two SQS structures with 27 atoms (6 Ge, 6 Sb, and 15 Te atoms) and 45 atoms (10 Ge, 10 Sb, and 25 Te atoms) in the unit cell and compared their electronic and transport properties. Despite the relatively poor quality of PCFs for the case with 27-atomic unit cell, we find that the overall calculated properties for these structures are very similar (a detailed comparison of electronic, optical, and transport properties in these structures can be found in Ref. 28). This may indicate that

the inherent properties in this phase are independent of its PCFs. Thus, this 45-atomic unit cell is converged with respect to size to calculate properties of *c*-GST.

The self-consistent calculations were performed with a $4 \times 4 \times 4$ k-point grid of 36 k points in the irreducible Brillouin zone (IBZ), whereas electronic and optical properties were calculated using an $8 \times 8 \times 8$ k-mesh. For the calculation of the thermoelectric properties, we employed a denser k-point mesh of $14 \times 14 \times 14$. All the calculations were duly tested for convergence with a variety of k-point grids. The thermopower was analysed using the Boltztrap code²⁹ employing Boltzmann transport theory within the constant scattering time approximation (CSTA). This CSTA approach has been shown to be successful in calculating the Seebeck coefficient in a variety of thermoelectric materials.^{30–34} The substance of CSTA is to assume that the energy dependence of the scattering rate is small compared with the energy dependence of the electronic structure. A detailed description of this approach can be found elsewhere.^{31,35} We used well converged basis sets with LAPW basis size corresponding to $R_{MT}K_{max}=9.0$, where R_{MT} is the smallest Muffin Tin radius and K_{max} is the plane-wave cutoff parameter. The LAPW sphere radius of 2.47 Bohr was used for Ge, whereas 2.5 Bohr was used for Sb and Te. Spin-orbit coupling was included for all the properties reported.

The total density of states (DOS) of *c*-GST near the Fermi energy (E_F) is shown in Fig. 1. We find a clear semi-conducting behavior with a band gap of 0.17 eV. This is in good agreement with a previous theoretical investigation,³⁶ where an indirect band gap of 0.1 eV was noted for *c*-GST although they used different exchange-correlation potential (local density approximation without spin-orbit coupling) and different method (shifting of the hexagonal structure) to generate the cubic structure. Another previous study considered the cubic GST as a narrow-gap degenerate semiconductor, with E_F reside inside the valence band or a defect band.³⁷ The experimentally reported estimated optical gap is ~ 0.5 eV (Refs. 37 and 38), which is larger than what we obtain.

GST has applications in rewritable DVD and Blu-ray technologies, where a large contrast between the optical properties of the cubic and amorphous phases is exploited.^{5,7}

FIG. 1. Calculated total density of states for *c*-GST. The Fermi level is set to zero.

FIG. 2. Calculated reflectivity of *c*-GST. Previous experimental result from Ref. 39 is also shown for a better comparison.

In Figs. 2 and 3, we present the averaged reflectivity and optical absorption coefficients for *c*-GST, respectively. It is evident from Fig. 2 that *c*-GST displays a high reflectivity ranging from 0.65 to 0.68 within the energy up to 3.5 eV, which covers the energy range of DVD (1.91 eV) and blu-ray (3.06 eV)⁵ technology. This is consistent with the measured optical reflectivity by García-García *et al.*³⁹ [Fig. 2], though they used thin films of *c*-GST in their experiments. Fig. 3 depicts the calculated optical absorption coefficients together with that from the previous measurements. Our calculated absorption coefficient compares well in magnitude with that from Lee *et al.*³⁷ and Pirovano *et al.*³⁸ for the higher energy regime but disagrees at lower energy reflecting the band gap difference. One possible explanation is that the structure is more complex than the random structure that we assumed, or that larger length scale effects such as local near ferroelectric polarizations play a role in the samples. Additionally, the dimension of the crystals may also play a role since thin films of GST were used in both of the experiments instead of the ideal bulk that we are considering.

In Fig. 4, we have plotted the calculated thermopower ($S(T)$) of *c*-GST at various temperatures ranging from 100 K up to 500 K, for both electron-doped (*n*-type) and hole-doped (*p*-type) cases. It has been reported that it might be possible to dope GST *n*-type,^{40,41} however, keeping in mind that most

FIG. 3. Calculated optical absorption coefficient along with experimental data from previous measurements (see Refs. 37 and 38).

FIG. 4. Calculated thermopowers of *c*-GST as a function of doping concentration both for *p*-type and *n*-type at various temperatures. Solid horizontal line at 200 $\mu\text{V/K}$ represents the limitation of Seebeck coefficients for good TE materials.

of the experiments deal with *p*-type GST, we focus on the thermoelectric properties of *p*-type *c*-GST. In fact, there is a rather good electron-hole symmetry in the $S(T)$, so *n*-type if realized would be similar to *p*-type. We give the direction average of $S(T)$, i.e., $S = \frac{S_{xx} + S_{yy} + S_{zz}}{3}$, to eliminate artificial anisotropy due to the unit supercell.

We find *c*-GST exhibits a Seebeck coefficients 200–323 $\mu\text{V/K}$ when doped *p*-type for doping concentration 1.7×10^{18} – $6.1 \times 10^{19} \text{ cm}^{-3}$ at room temperature. More importantly, at low temperature ($T = 200 \text{ K}$), the Seebeck coefficient exceeds 200 $\mu\text{V/K}$ for a doping range of 1×10^{19} – $3.5 \times 10^{19} \text{ cm}^{-3}$. A strong bipolar effect was noted for $T = 300 \text{ K}$ and higher temperatures, however, it was not present at lower temperatures. This is connected with the band gap, which as noted may be too low in the present calculations. In order to assess the possibility of *c*-GST as a low temperature TE material, it is necessary to compare the calculated Seebeck coefficient with that in popular low-temperature TE materials. For example, Poudel and co-workers³ reported a bulk nano-composite $\text{Bi}_x\text{Sb}_{2-x}\text{Te}_3$ with $S(T)$ of 185 $\mu\text{V/K}$ at 300 K. Chung *et al.*⁴ discovered CsBi_4Te_6 with a thermopower of 105 $\mu\text{V/K}$ at 300 K. Importantly, the high thermopowers are at high carrier concentrations where reasonable electric conductivity may be anticipated.

We shall now focus on comparing our calculated results with previous experiments on GST thin films. Kato and Tanaka¹⁷ investigated thermoelectric properties of $\text{Ge}_2\text{Sb}_2\text{Te}_5$ thin films with sample dimensions ranging from 50 nm to 600 nm for a temperature range of 25 °C–200 °C. For their *c*-GST samples, they found a pretty high values of $S(T)$ from 280 $\mu\text{V/K}$ to 380 $\mu\text{V/K}$. Lee *et al.*¹¹ also measured the temperature dependence of $S(T)$ with 25 nm and 125 nm $\text{Ge}_2\text{Sb}_2\text{Te}_5$ thin films from 20 °C to 300 °C and observed $S(T)$ as high as 205 $\mu\text{V/K}$ and 175 $\mu\text{V/K}$ for 125 nm and 25 nm thick films, respectively. Both of the authors found a decreasing $S(T)$ with increasing temperatures. In order to further compare our calculations with these measurements, we matched the calculated Seebeck coefficients at 300 K with that from Kato's data for the 550 nm thick thin film and calculated the corresponding doping levels in *c*-GST. For these particular doping levels, we then plotted the variation of $S(T)$ with temperature together with the

FIG. 5. Temperature dependence of the thermopower of *c*-GST with two *p*-type doping concentrations and experimental results from previous work.^{11,17}

respective experimental data, as shown in Fig. 5. As seen, a similar decreasing trend of $S(T)$ with respect to temperatures can be found in the calculation for both doping concentrations. However, the agreement of the decreasing feature is more obvious in the low doping level case, which falls into the bipolar region and the deviation may stem from the underestimation of the band gap. This may also suggest the fact that the experimental samples are in the low carrier concentrations range, where bipolar effect is strong and detrimental to the thermoelectric properties. Therefore, a better thermoelectric performance might be feasible in higher doping concentration samples.

To summarize, we performed the first-principles calculations to investigate electronic, optical, and thermoelectric properties of *c*-GST with fully random cation site occupancy generated using the SQS scheme. Our results based on Boltzmann transport calculations predict substantially high thermopower at high carrier concentration ($\sim 300 \mu\text{V/K}$ at $7 \times 10^{18} \text{ cm}^{-3}$) at room temperature. We also find high thermopower at high carrier concentrations going down to 200 K and below suggesting that with heavy hole doping *c*-GST may be an excellent low-temperature thermoelectric. These results suggest future experiments to better understand the low-temperature thermoelectric properties of *c*-GST. Specifically, it would be of interest to study the temperature range of 200 K–300 K with high doping concentrations ranging from $1 \times 10^{19} \text{ cm}^{-3}$ up to $6 \times 10^{19} \text{ cm}^{-3}$.

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